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=> s (perfluoropolyether or polyether or polyoxyalkylene ether) and (gas? (2a) hydrogen or reduc? or hydrogenation) and (group viii or palladium or platinum or rhodium or ruthenium or pd or pt or rh or ru) and (support? (5a) fluoride)

9 FILES SEARCHED...

14 FILES SEARCHED...

22 FILES SEARCHED...

28 FILES SEARCHED...

36 FILES SEARCHED...

41 FILES SEARCHED...

49 FILES SEARCHED...

52 FILES SEARCHED...

58 FILES SEARCHED...

62 FILES SEARCHED...

64 FILES SEARCHED...

67 FILES SEARCHED...

75 FILES SEARCHED...

L1 87 (PERFLUOROPOLYETHER OR POLYETHER OR POLYOXYALKYLENE ETHER) AND  
(GAS? (2A) HYDROGEN OR REDUC? OR HYDROGENATION) AND (GROUP VIII  
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RH OR RU) AND (SUPPORT? (5A) FLUORIDE)

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=> d l2 1-82 ti

L2 ANSWER 1 OF 82 PROMT COPYRIGHT 2004 Gale Group on STN

TI Trade name directory.

L2 ANSWER 2 OF 82 CAPLUS COPYRIGHT 2004 ACS on STN

TI Process for the preparation of perfluoropolyethers having aldehyde,  
alcohol, and amine end groups by catalytic reduction

L2 ANSWER 3 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN

TIEN Azeotropic compositions comprising 1,1,1,2,3,3,3-heptafluoropropane and  
processes using said compositions.

L2 ANSWER 4 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN

TIEN PROTON-CONDUCTIVE POLYMER FILM AND PROCESS FOR PRODUCING THE SAME.

L2 ANSWER 5 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN

TIEN PROCESSES FOR THE PRODUCTION OF HEXAFLUOROPROPENE AND OPTIONALLY OTHER  
HALOGENATED HYDROCARBONS CONTAINING FLUORINE.

L2 ANSWER 6 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN

TIEN PROCESSES FOR THE MANUFACTURE OF 1,1,1,3,3-PENTAFLUOROPROPENE,  
2-CHLORO-PENTAFLUOROPROPENE.

L2 ANSWER 7 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN

10/631,862

TIEN ELECTROSTATIC PROCESSING OF ELECTROCHEMICAL DEVICE COMPONENTS  
TIFR TRAITEMENT ELECTROSTATIQUE DE COMPOSANTS DE DISPOSITIFS ELECTROCHIMIQUES

L2 ANSWER 8 OF 82 USPATFULL on STN

TI Fuel cell, fuel cell generator, and equipment using the same

L2 ANSWER 9 OF 82 USPATFULL on STN

TI Proton-conductive polymer film and process for producing the same

L2 ANSWER 10 OF 82 USPATFULL on STN

TI Reagents and methods for library synthesis and screening

L2 ANSWER 11 OF 82 PROMT COPYRIGHT 2004 Gale Group on STN

TI Trade name directory. (A-O).

L2 ANSWER 12 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN

TIEN Polymerization of cyclic ethers using heterogeneous catalysts.

TIEN Polymerization of cyclic ethers using heterogeneous catalysts.

L2 ANSWER 13 OF 82 USPATFULL on STN

DUPLICATE 1

TI Production of 1,2-dihydro and 2,2-dihydro hexafluoropropanes and azeotropes thereof with HF

L2 ANSWER 14 OF 82 USPATFULL on STN

DUPLICATE 2

TI Process for the manufacture of 1,1,1,3,3-pentafluoropropene, 2-chloro-pentafluoropropene and compositions comprising saturated derivatives

L2 ANSWER 15 OF 82 USPATFULL on STN

DUPLICATE 3

TI PROCESSES FOR THE MANUFACTURE OF 1,1,1,3,3- PENTAFLUOROPROPENE, 2-CHLORO-PENTAFLUOROPROPENE AND COMPOSITIONS COMPRISING SATURATED DERIVATIVES THEREOF

L2 ANSWER 16 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN

TIEN Fuel cell, fuel cell generator, and equipment using the same.

L2 ANSWER 17 OF 82 USPATFULL on STN

TI Fuel cell, fuel cell generator, and equipment using the same

L2 ANSWER 18 OF 82 USPATFULL on STN

TI Interfacially polymerized, bipiperidine-polyamide membranes for reverse osmosis and/or nanofiltration and process for making the same

L2 ANSWER 19 OF 82 USPATFULL on STN

TI Production of 1,2-dihydro and 2,2-dihydro hexafluoropropanes and azeotropes thereof with HF

L2 ANSWER 20 OF 82 USPATFULL on STN

DUPLICATE 4

TI Fuel cell with monolithic flow field-bipolar plate assembly and method for making and cooling a fuel cell stack

L2 ANSWER 21 OF 82 USPATFULL on STN

DUPLICATE 5

TI Gas diffusion electrode with nanosized pores and method for making same

L2 ANSWER 22 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN

TIEN GAS DIFFUSION ELECTRODE WITH NANOSIZED PORES AND METHOD FOR MAKING SAME

TIFR ELECTRODE DE DIFFUSION GAZEUSE A PORES DE TAILLE NANOMETRIQUE ET PROCEDE POUR LA FABRICATION D'UNE TELLE ELECTRODE

L2 ANSWER 23 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN

TIEN FUEL CELL WITH MONOLITHIC FLOW FIELD-BIPOLAR PLATE ASSEMBLY AND METHOD

- FOR MAKING AND COOLING A FUEL CELL STACK
- TIFR PILE A COMBUSTIBLE A ASSEMBLAGE DE PLAQUES A CHAMP BIPOLAIRE ET ECOULEMENT MONOLITHIQUE, ET PROCEDE DE FABRICATION ET DE REFROIDISSEMENT D'UN EMPILEMENT DE PILES A COMBUSTIBLE
- L2 ANSWER 24 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN
- TIEN ELECTRONICALLY CONDUCTING FUEL CELL COMPONENT WITH DIRECTLY BONDED LAYERS AND METHOD FOR MAKING SAME
- TIFR COMPOSANT DE PILE A COMBUSTIBLE CONDUCTEUR SUR LE PLAN ELECTRONIQUE DOTE DE COUCHES DIRECTEMENT LIEES ET PROCEDE DE FABRICATION CORRESPONDANT
- L2 ANSWER 25 OF 82 USPATFULL on STN
- TI Electronically conducting fuel cell component with directly bonded layers and method for making same
- L2 ANSWER 26 OF 82 USPATFULL on STN
- TI Processes for the production of hexafluoropropene and optionally other halogenated hydrocarbons containing fluorine
- L2 ANSWER 27 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN
- TIEN Interfacially polymerized, bipiperidine-polyamide membranes for reverse osmosis and/or nanofiltration and process for making the same.
- L2 ANSWER 28 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN
- TIEN PRODUCTION OF 1,2-DIHYDRO AND 2,2-DIHYDRO HEXAFLUOROPROPANES AND AZEOTROPES THEREOF WITH HF.
- L2 ANSWER 29 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN
- TIEN Interfacially synthesized reverse osmosis membranes and processes for preparing the same.
- TIEN Interfacially synthesized reverse osmosis membranes and processes for preparing the same.
- L2 ANSWER 30 OF 82 USPATFULL on STN
- TI Catalysts for halogenated hydrocarbon processing, their precursors and their preparation and use
- L2 ANSWER 31 OF 82 USPATFULL on STN
- TI Process for the manufacture of 2-chloro-1,1,1-trifluoroethane
- L2 ANSWER 32 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN
- TIEN PROCESSES FOR THE PRODUCTION OF HEXAFLUOROPROPENE AND OPTIONALLY OTHER HALOGENATED HYDROCARBONS CONTAINING FLUORINE
- TIFR PROCEDES RELATIFS A LA PRODUCTION D'HEXAFLUOROPROPENE ET EVENTUELLEMENT D'AUTRES HYDROCARBURES HALOGENES CONTENANT DU FLUOR
- L2 ANSWER 33 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN
- TIEN SELECTIVE MEMBRANE AND PROCESS FOR ITS PREPARATION
- TIFR MEMBRANE SELECTIVE ET PROCEDE DE PREPARATION DE CELLE-CI
- L2 ANSWER 34 OF 82 USPATFULL on STN
- TI Catalytic halogenated hydrocarbon processing and ruthenium catalysts for use therein
- L2 ANSWER 35 OF 82 USPATFULL on STN
- TI Process for the production of trifluoroethylene
- L2 ANSWER 36 OF 82 USPATFULL on STN
- TI Polymerization of, and depolymerization to, cyclic ethers using selected metal compound catalysts
- L2 ANSWER 37 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN



- TIEN CATALYSTS FOR HALOGENATED HYDROCARBON PROCESSING, THEIR PRECURSORS AND THEIR PREPARATION AND USE
- TIFR CATALYSEURS DE TRAITEMENT D'HYDROCARBURES HALOGENES, LEURS PRECURSEURS, LEUR PREPARATION ET LEUR UTILISATION
- L2 ANSWER 38 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN
- TIEN CATALYTIC HALOGENATED HYDROCARBON PROCESSING AND RUTHENIUM CATALYSTS FOR USE THEREIN
- TIFR TRAITEMENT PAR CATALYSE DES HYDROCARBURES HALOGENES ET CATALYSEURS AU RUTHENIUM UTILISES
- L2 ANSWER 39 OF 82 USPATFULL on STN
- TI Polymerization of, and depolymerization to, cyclic ethers using selected metal compound catalysts
- L2 ANSWER 40 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN
- TIEN PROCESS FOR THE MANUFACTURE OF 2-CHLORO-1,1,1-TRIFLUOROETHANE.
- L2 ANSWER 41 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN
- TIEN PROCESS FOR THE MANUFACTURE OF 2-CHLORO-1,1,1,2-TETRAFLUOROETHANE AND PENTAFLUOROETHANE.
- L2 ANSWER 42 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN
- TIEN PROCESS FOR THE MANUFACTURE OF 1,1,1,2-TETRAFLUOROETHANE.
- L2 ANSWER 43 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN
- TIEN PROCESS FOR THE MANUFACTURE OF 2,2-DICHLORO-1,1,1-TRIFLUOROETHANE, 2-CHLORO-1,1,1,2-TETRAFLUOROETHANE AND PENTAFLUOROETHANE.
- L2 ANSWER 44 OF 82 USPATFULL on STN
- TI Production of 1,2-dihydro and 2,2-dihydro hexafluoropropanes and azeotropes thereof with HF
- L2 ANSWER 45 OF 82 USPATFULL on STN
- TI Acid gas fractionation process
- L2 ANSWER 46 OF 82 USPATFULL on STN
- TI Acid gas fractionation process for fossil fuel gasifiers
- L2 ANSWER 47 OF 82 USPATFULL on STN
- TI Process for manufacture of high purity 1, 1-dichlorotetrafluoroethane
- L2 ANSWER 48 OF 82 USPATFULL on STN
- TI Polymerization of, and depolymerization to, cyclic ethers using selected metal compound catalysts
- L2 ANSWER 49 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN
- TIEN REGENERATION OR ACTIVATION OF NOBLE METAL CATALYSTS USING FLUOROHALOCARBONS OR FLUOROHALOHYDROCARBONS.
- L2 ANSWER 50 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN
- TIEN MANUFACTURE OF 1,1,1,2-TETRAFLUOROETHANE.
- L2 ANSWER 51 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN
- TIEN PRODUCTION OF 1,2-DIHYDRO AND 2,2-DIHYDRO HEXAFLUOROPROPANES AND AZEOTROPES THEREOF WITH HF
- TIFR PRODUCTION DE 1,2-DIHYDRO ET 2,2-DIHYDRO HEXAFLUOROPROPANES ET D'AZEOTROPES DE CES DERNIERS A L'AIDE DE FLUORURE D'HYDROGENE
- L2 ANSWER 52 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN
- TIEN PROCESS FOR MANUFACTURE OF HIGH PURITY 1,1-DICHLOROTETRAFLUOROETHANE
- TIFR PRODECE POUR PRODUIRE DU 1,1-DICHLOROTETRAFLUOROETHANE HAUTEMENT PUR

- L2 ANSWER 53 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN MEMBRANE AND NON-MEMBRANE SOUR GAS TREATMENT PROCESS  
 TIFR PROCEDE AVEC ET SANS MEMBRANE DE TRAITEMENT DE GAZ SULFUREUX
- L2 ANSWER 54 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN SOUR GAS TREATMENT PROCESS  
 TIFR PROCEDE DE TRAITEMENT DE GAZ SULFUREUX
- L2 ANSWER 55 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN SOUR GAS MEMBRANE TREATMENT PROCESS INCLUDING DEHYDRATION  
 TIFR PROCEDE DE TRAITEMENT MEMBRANAIRE DE GAZ SULFUREUX INCLUANT LA  
 DESHYDRATATION
- L2 ANSWER 56 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN POLYMERIZATION, AND DEPOLYMERIZATION, OF CYCLIC ETHERS USING  
 HETEROGENEOUS CATALYSTS  
 TIFR POLYMERISATION ET DEPOLYMERISATION D'ETHERS CYCLIQUES A L'AIDE DE  
 CATALYSEURS HETEROGENES
- L2 ANSWER 57 OF 82 USPATFULL on STN  
 TI Process for manufacture of high purity 1,1-dichlorotetrafluoroethane
- L2 ANSWER 58 OF 82 USPATFULL on STN  
 TI Sour gas treatment process
- L2 ANSWER 59 OF 82 USPATFULL on STN  
 TI Sour gas treatment process including membrane and non-membrane treatment  
 steps
- L2 ANSWER 60 OF 82 USPATFULL on STN  
 TI Sour gas treatment process including dehydration of the gas stream
- L2 ANSWER 61 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN  
 TIEN ACTIVATION OF NOBLE METAL CATALYSTS FOR USE IN HYDRODEHALOGENATION OF  
 HALOGEN-SUBSTITUTED HYDROCARBONS CONTAINING FLUORINE AND AT LEAST ONE  
 OTHER HALOGEN.
- L2 ANSWER 62 OF 82 USPATFULL on STN  
 TI Manufacture of 1,1,1,2-tetrafluoroethane
- L2 ANSWER 63 OF 82 USPATFULL on STN  
 TI Process for the manufacture of 1,1,1,2-tetrafluoroethane
- L2 ANSWER 64 OF 82 USPATFULL on STN  
 TI Process for the manufacture of 2,2-dichloro-1,1,1-trifluoroethane,  
 2-chloro-1,1,1,2-tetrafluoroethane and pentafluoroethane
- L2 ANSWER 65 OF 82 USPATFULL on STN  
 TI Process for the manufacture of 2-chloro-1,1,1,2-tetrafluoroethane and  
 pentafluoroethane
- L2 ANSWER 66 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN  
 TIEN Gas-phase fluorination process.  
 TIEN Gas-phase fluorination process.
- L2 ANSWER 67 OF 82 USPATFULL on STN  
 TI Interfacially synthesized reverse osmosis membranes and processes for  
 preparing the same
- L2 ANSWER 68 OF 82 USPATFULL on STN  
 TI Activation of noble metal catalysts for use in hydrodehalogenation of

halogen-substituted hydrocarbons containing fluorine and at least one other halogen

- L2 ANSWER 69 OF 82 USPATFULL on STN  
 TI Manufacture of 1,1,1,2-tetrafluoroethane
- L2 ANSWER 70 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN PROCESS FOR REMOVING CONDENSABLE COMPONENTS FROM GAS STREAMS  
 TIFR PROCEDE SERVANT A RETIRER DE FLUX GAZEUX DES CONSTITUANTS CONDENSABLES
- L2 ANSWER 71 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN PROCESS FOR THE MANUFACTURE OF 2-CHLORO-1,1,1,2-TETRAFLUOROETHANE AND PENTAFLUOROETHANE  
 TIFR PROCEDE DE FABRICATION DE 2-CHLORO-1,1,1,2-TETRAFLUOROETHANE ET PENTAFLUOROETHANE
- L2 ANSWER 72 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN PROCESS FOR THE MANUFACTURE OF 1,1,1,2-TETRAFLUOROETHANE  
 TIFR PROCEDE DE FABRICATION DE 1,1,1,2-TETRAFLUOROETHANE
- L2 ANSWER 73 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN PROCESS FOR THE MANUFACTURE OF 2-CHLORO-1,1,1-TRIFLUOROETHANE  
 TIFR PROCEDE DE FABRICATION DE 2-CHLORO-1,1,1-TRIFLUOROETHANE
- L2 ANSWER 74 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN PROCESS FOR THE MANUFACTURE OF 2,2-DICHLORO-1,1,1-TRIFLUOROETHANE, 2-CHLORO-1,1,1,2-TETRAFLUOROETHANE AND PENTAFLUOROETHANE  
 TIFR PROCEDE DE FABRICATION DE 2,2-DICHLORO-1,1,1-TRIFLUOROETHANE, 2-CHLORO-1,1,1,2-TETRAFLUOROETHANE ET PENTAFLUOROETHANE
- L2 ANSWER 75 OF 82 USPATFULL on STN  
 TI Activation of noble metal catalysts using fluorohalocarbons or fluorohalohydrocarbons
- L2 ANSWER 76 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN ACTIVATION OF NOBLE METAL CATALYSTS FOR USE IN HYDRODEHALOGENATION OF HALOGEN-SUBSTITUTED HYDROCARBONS CONTAINING FLUORINE AND AT LEAST ONE OTHER HALOGEN  
 TIFR ACTIVATION DE CATALYSEURS DE METAUX PRECIEUX DESTINES A L'HYDRODESHALOGENATION DES HYDROCARBURES SUBSTITUES PAR HALOGENE ET CONTENANT DU FLUOR ET AU MOINS UN AUTRE HALOGENE
- L2 ANSWER 77 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN REGENERATION OR ACTIVATION OF NOBLE METAL CATALYSTS USING FLUOROHALOCARBONS OR FLUOROHALOHYDROCARBONS  
 TIFR REGENERATION OU ACTIVATION D'UN CATALYSEUR EN METAL PRECIEUX A L'AIDE D'HALOCARBONES FLUORES OU D'HALOHYDROCARBONES FLUORES
- L2 ANSWER 78 OF 82 USPATFULL on STN  
 TI Regeneration of noble metal catalysts used in hydrodehalogenation of halogen-substituted hydrocarbons containing fluorine and at least one other halogen
- L2 ANSWER 79 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN  
 TIEN MANUFACTURE OF 1,1,1,2-TETRAFLUOROETHANE  
 TIFR FABRICATION DE 1,1,1,2-TETRAFLUOROETHANE
- L2 ANSWER 80 OF 82 USPATFULL on STN  
 TI Regeneration or activation of noble metal catalysts using fluorohalocarbons or fluorohalohydrocarbons
- L2 ANSWER 81 OF 82 USPATFULL on STN

10/631,862

TI Gas-phase fluorination process

L2 ANSWER 82 OF 82 JAPIO (C) 2004 JPO on STN

TI METHOD FOR PRODUCING PERFLUOROPOLYETHERS HAVING ALDEHYDE, ALCOHOL, AND AMINE TERMINAL GROUPS

=> d 2,5,26,30,34,37,81,82 bib ab

L2 ANSWER 2 OF 82 CAPLUS COPYRIGHT 2004 ACS on STN

AN 2004:117259 CAPLUS

DN 140:146686

TI Process for the preparation of perfluoropolyethers having aldehyde, alcohol, and amine end groups by catalytic reduction

IN Di, Meo Antonello; Picozzi, Rosaldo; Tonelli, Claudio

PA Solvay Solexis S.P.A., Italy

SO Eur. Pat. Appl., 10 pp.

CODEN: EPXXDW

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	EP 1388556	A2	20040211	EP 2003-17183	20030729
	EP 1388556	A3	20040331		
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK				
	US 2004068144	A1	20040408	US 2003-630698	20030731
	JP 2004068007	A2	20040304	JP 2003-205414	20030801
PRAI	IT 2002-MI1734	A	20020801		

AB A process for the perfluoropolyether preparation having reactive end groups -CH<sub>2</sub>NH<sub>2</sub>, -CHO, -CH<sub>2</sub>OH, by reduction of the corresponding perfluoropolyethers having -CN, -COCl, -CHO end groups by using gaseous hydrogen in the presence of a catalyst constituted by Pd, Rh, or Ru, supported on solid metal fluorides, at 20-150° and under a pressure between 1 and 50 atmospheric is disclosed.

L2 ANSWER 5 OF 82 EUROPATFULL COPYRIGHT 2004 WILA on STN

GRANTED PATENT - ERTEILTES PATENT - BREVET DELIVRE

AN 1084093 EUROPATFULL ED 20040819 EW 200434 FS PS

TIEN PROCESSES FOR THE PRODUCTION OF HEXAFLUOROPROPENE AND OPTIONALLY OTHER HALOGENATED HYDROCARBONS CONTAINING FLUORINE.

TIDE VERFAHREN ZUR HERSTELLUNG VON HEXAFLUORPROPEN UND GEGEBENENFALLS WEITEREN HALOGENIERTEN FLUOR ENTHALTENDEN KOHLENWASSERSTOFFEN.

TIFR PROCEDES RELATIFS A LA PRODUCTION D'HEXAFLUOROPROPENE ET EVENTUELLEMENT D'AUTRES HYDROCARBURES HALOGENES CONTENANT DU FLUOR.

IN SIEVERT, Allen, Capron, 215 Rhett Lane, Elkton, MD 21921, US;  
RAO, Velliyur, Nott, Mallikarjuna, 1 Georgetown Avenue, Wilmington, DE 19809, US;

PA WALCZAK, Francis, J., 203 Jefferson Avenue, New Castle, DE 19720, US  
E.I. DUPONT DE NEMOURS AND COMPANY, 1007 Market Street, Wilmington, Delaware 19898, US

PAN 2567250

AG Towler, Philip Dean et al., Frank B. Dehn & Co., European Patent Attorneys, 179 Queen Victoria Street, London EC4V 4EL, GB

AGN 75321

OS MEPB2004035 EP 1084093 B1 0015

SO Wila-EPS-2004-H34-T1

DT Patent

10/631,862

LA Anmeldung in Englisch; Veroeffentlichung in Englisch  
DS R DE; R FR; R GB; R IT; R NL  
PIT EPB1 EUROPÄISCHE PATENTSCHRIFT (Internationale Anmeldung)  
PI EP 1084093 B1 20040818  
OD 20010321  
AI EP 1999-928367 19990602  
PRAI US 1998-87751 19980602  
RLI WO 99-US12246 990602 INTAKZ  
WO 1999062851 991209 INTPNR  
REP EP 434407 A EP 434409 A  
WO 90-08748 A WO 97-19751 A  
GB 821211 A GB 2313118 A  
US 2576823 A US 5523501 A

L2 ANSWER 26 OF 82 USPATFULL on STN  
AN 2001:226805 USPATFULL  
TI Processes for the production of hexafluoropropene and optionally other  
halogenated hydrocarbons containing fluorine  
IN Sievert, Allen Capron, Elkton, MD, United States  
Rao, V. N. Mallikarjuna, Wilmington, DE, United States  
Walczak, Francis J., New Castle, DE, United States  
PA E. I. du Pont de Nemours and Company, Wilmington, DE, United States  
(U.S. corporation)  
PI US 6329559 B1 20011211  
WO 9962851 19991209  
AI US 2000-701448 20001127 (9)  
WO 1999-US12246 19990602  
20001127 PCT 371 date  
20001127 PCT 102(e) date  
PRAI US 1998-87751P 19980602 (60)  
DT Utility  
FS GRANTED  
EXNAM Primary Examiner: Siegel, Alan  
CLMN Number of Claims: 20  
ECL Exemplary Claim: 1  
DRWN 1 Drawing Figure(s); 1 Drawing Page(s)  
LN.CNT 961  
CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB A process is disclosed for the manufacture of CF.sub.3 CF.dbd.CF.sub.2,  
and optionally a least one compound selected from CF.sub.3 CH.sub.2  
CF.sub.3 and CF.sub.3 CHFCHF.sub.2. The process involves contacting a  
reactor feed including a precursor stream of at least one halogenated  
propane of the formula CX.sub.3 CH.sub.2 CH.sub.y X.sub.(3-y) and/or  
halogenated propene of the formula CX.sub.3 CH.dbd.CH.sub.y X.sub.(2-y),  
where each X is Cl or F and y is 0, 1 or 2 (provided that the average  
fluorine content of the precursor stream is no more than 5 fluorine  
substituents per molecule) with HF and Cl.sub.2 in a chlorofluorination  
reaction zone containing a fluorination catalyst and operating at a  
temperature between about 150° C. and 400° C., to produce  
a reaction zone effluent including HF, HCl and a mixture of reaction  
products of the precursor feed which contains at least one compound of  
the formula C.sub.3 Cl.sub.2 F.sub.6 including CClF.sub.2 CClFCF.sub.3  
and at least one compound of the formula C.sub.3 HClF.sub.6, including  
CHF.sub.2 CClFCF.sub.3 and has an average fluorine content which is at  
least one fluorine substituent per molecule more than the average  
fluorine content of the precursor stream. The chlorofluorination  
reaction zone effluent is distilled to produce (i) a low-boiling  
component including HCl (and when they are present in the reaction zone  
effluent, C.sub.3 F.sub.8, C.sub.3 ClF.sub.7 and C.sub.3 HF.sub.7), (ii)  
a hydrogenation feed component containing at least one compo  
of the formula C.sub.3 Cl.sub.2 F.sub.6 including CClF.sub.2  
CClFCF.sub.3 and at least one compound of the formnula C.sub.3

HClF.sub.6 including CHF.sub.2 CClFCF.sub.3, and an underfluorinated component including halogenated propanes containing at least one chlorine substituent and from one to five fluorine substituents. The CClF.sub.2 CClFCF.sub.3 and CHF.sub.2 CClFCF.sub.3 of hydrogenation feed component (ii) is reacted with hydrogen to produce a mixture including CF.sub.3 CF.dbd.CF.sub.2 and CF.sub.3 CHFCHF.sub.2 and the CF.sub.3 CF.dbd.CF.sub.2 from this product mixture is recovered. Underfluorinated component (iii) is returned to the chlorofluorination reaction zone.

L2 ANSWER 30 OF 82 USPATFULL on STN

AN 2000:132057 USPATFULL

TI Catalysts for halogenated hydrocarbon processing, their precursors and their preparation and use

IN Duzick, Timothy C., Hockessin, DE, United States

Rao, Velliyur Nott Mallikarjuna, Wilmington, DE, United States

Subramanian, Munirpallam A., Kennett Square, PA, United States

PA E. I. du Pont de Nemours and Company, Wilmington, DE, United States (U.S. corporation)

PI US 6127585 20001003

WO 9719751 19970605

AI US 1998-77267 19980527 (9)

WO 1996-US18967 19961126

19980527 PCT 371 date

19980527 PCT 102(e) date

PRAI US 1995-7734P 19951129 (60)

DT Utility

FS Granted

EXNAM Primary Examiner: Wu, David W.; Assistant Examiner: Zalukaeva, Tanya

CLMN Number of Claims: 20

ECL Exemplary Claim: 1

DRWN No Drawings

LN.CNT 958

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB Processes are disclosed for decreasing the chlorine to carbon ratio for halogenated hydrocarbons containing chlorine and from 1 to 6 carbon atoms, in the presence of a multiphase catalyst. The processes each involve (1) preparing a single phase solid catalyst precursor which has a structure that collapses at a temperature of about 400° C. or less and has the formula (NH.sub.3).sub.6 Ru.sub.1-r-s Co.sub.r Cr.sub.s MF.sub.6, where r+s is in the range of 0.00 to 0.99, and M is at least one trivalent metal selected from the group consisting of Al, Cr, Fe, V, Sc and Ga; and (2) producing the multiphase catalyst by heating the single phase solid catalyst precursor to about 400° C. or less in an non-oxidizing atmosphere to produce a multiphase composition wherein a phase containing ruthenium is homogeneously dispersed with a phase containing metal fluoride.

Also disclosed are single phase fluoride compositions having the formula (NH.sub.3).sub.6 Ru.sub.1-r-s Co.sub.r Cr.sub.s MF.sub.6, where r+s is in the range of 0.00 to 0.99, and M is at least one trivalent element selected from the group consisting of Al, Cr, Fe, V, Sc and Ga; and multiphase catalyst compositions consisting essentially of metallic ruthenium and fluorides of at least one element selected from the group consisting of Al, Co, Cr, Fe, V, Sc and Ga, wherein the ruthenium is homogeneously dispersed with phases of the fluorides.

L2 ANSWER 34 OF 82 USPATFULL on STN

AN 1999:75850 USPATFULL

TI Catalytic halogenated hydrocarbon processing and ruthenium catalysts for use therein

10/631,862

IN Rao, Velliyur Nott Mallikarjuna, Wilmington, DE, United States  
PA E. I. du Pont de Nemours and Company, Wilmington, DE, United States  
(U.S. corporation)

PI US 5919994 19990706  
WO 9719750 19970605

AI US 1997-875470 19970728 (8)  
WO 1996-US18952 19961126  
19970728 PCT 371 date  
19970728 PCT 102(e) date

PRAI US 1995-7702P 19951129 (60)

DT Utility

FS Granted

EXNAM Primary Examiner: Yildirim, Bekir L.

CLMN Number of Claims: 8

ECL Exemplary Claim: 1

DRWN No Drawings

LN.CNT 1023

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB Processes for decreasing the chlorine to carbon ratio for halogenated hydrocarbons containing chlorine and from 1 to 6 carbon atoms, in the presence of a catalyst are disclosed. The processes are each characterized by employing a catalyst comprising ruthenium on a support of (i) fluorided alumina, (ii) aluminum fluoride, or (iii) fluorides of Zn, Mg, Ca, Ba, Y, Sm, Eu, and/or Dy. Also disclosed are multiphase catalyst compositions of ruthenium supported on fluorides of Zn, Mg, Ca, Ba, Y, Sm, Eu and/or Dy.

L2 ANSWER 37 OF 82 PCTFULL COPYRIGHT 2004 Univentio on STN

AN 1997019751 PCTFULL ED 20020514

TIEN CATALYSTS FOR HALOGENATED HYDROCARBON PROCESSING, THEIR PRECURSORS AND THEIR PREPARATION AND USE

TIFR CATALYSEURS DE TRAITEMENT D'HYDROCARBURES HALOGENES, LEURS PRECURSEURS, LEUR PREPARATION ET LEUR UTILISATION

IN DUZICK, Timothy, C.;  
RAO, Velliyur, Nott, Mallikarjuna;  
SUBRAMANIAN, Munirpallam, A.

PA E.I. DU PONT DE NEMOURS AND COMPANY;  
DUZICK, Timothy, C.;  
RAO, Velliyur, Nott, Mallikarjuna;  
SUBRAMANIAN, Munirpallam, A.

LA English

DT Patent

PI WO 9719751 A1 19970605

DS W: JP US AT BE CH DE DK ES FI FR GB GR IE IT LU MC NL PT SE

AI WO 1996-US18967 A 19961126

PRAI US 1995-60/007,734 19951129

ABEN Processes are disclosed for decreasing the chlorine to carbon ratio for halogenated hydrocarbons containing chlorine and from 1 to 6 carbon atoms, in the presence of a multiphase catalyst. The processes each involve (1) preparing a single phase solid catalyst precursor which has a structure that collapses at a temperature of about 400 °C or less and has the formula  $(\text{NH}_3)_6\text{Ru}_{1-r-s}\text{Cr}_r\text{Cr}_s\text{MF}_6$ , where  $r+s$  is in the range of 0.00 to 0.99, and M is at least one trivalent metal selected from the group consisting of Al, Cr, Fe, V, Sc and Ga; and (2) producing the multiphase catalyst by heating the single phase solid catalyst precursor to about 400 °C or less in a non-oxidizing atmosphere to produce a multiphase composition wherein a phase containing ruthenium

is homogeneously dispersed with a phase containing metal fluoride. Also disclosed are single phase fluoride compositions having the formula  $(\text{NH}_3)_6\text{Ru}_{1-r}\text{Sc}_r\text{Cr}_s\text{MF}_6$ , where  $r+s$  is in the range of 0.00 to 0.99, and M is at least one trivalent element selected from the group consisting of Al, Cr, Fe, V, Sc and Ga; and multiphase catalyst compositions consisting essentially of metallic ruthenium and fluorides of at least one element selected from the group consisting of Al, Co, Cr, Fe, V, Sc and Ga, wherein the ruthenium is homogeneously dispersed with phases of the fluorides.

ABFR L'invention concerne des procedes servant a diminuer le rapport entre le chlore et le carbone pour des hydrocarbures halogenes contenant du chlore et de 1 a 6 atomes de carbone, en presence d'un catalyseur a phases multiples. Ces procedes consistent chacun (1) a preparer un precurseur de catalyseur solide monophasé, dont la structure s'affaisse a une temperature egale ou inferieure a  $400^\circ\text{C}$  et qui possede la formule  $(\text{NH}_3)_6\text{Ru}_{1-r}\text{Sc}_r\text{Cr}_s\text{MF}_6$  dans laquelle  $r+s$  est situe dans la plage de 0,00 a 0,99 et M represente au moins un metal trivalent selectionne dans le groupe constitue par Al, Cr, Fe, V, Sc et Ga et (2) a produire le catalyseur a phases multiples par rechauffement du catalyseur solide monophasé a une temperature egale ou inferieure a  $400^\circ\text{C}$  dans une atmosphere non oxydante, afin d'obtenir une composition a phases multiples dans laquelle une phase contenant ruthenium est dispersee de facon homogene avec une phase contenant fluorure metallique. L'invention concerne egalement des compositions monophasées de fluor possedant la formule  $(\text{NH}_3)_6\text{Ru}_{1-r}\text{Sc}_r\text{Cr}_s\text{MF}_6$  dans laquelle  $r+s$  est situe dans la plage de 0,00 a 0,99 et M represente au moins un element trivalent selectionne dans le groupe constitue par Al, Cr, Fe, V, Sc et Ga, ainsi que des compositions de catalyseur a phases multiples constituees essentiellement par ruthenium et des fluorures metalliques d'au moins un element selectionne dans le groupe constitue par Al, Co, Cr, Fe, V, Sc et Ga, le ruthenium etant disperse de facon homogene avec des phases des fluorures.

L2 ANSWER 81 OF 82 USPATFULL on STN  
 AN 90:34289 USPATFULL  
 TI Gas-phase fluorination process  
 IN Manzer, Leo E., Wilmington, DE, United States  
 PA E. I. du Pont de Nemours and Company, Wilmington, DE, United States  
 (U.S. corporation)  
 PI US 4922037 19900501  
 AI US 1989-355867 19890519 (7)  
 RLI Continuation of Ser. No. US 1988-160003, filed on 24 Feb 1988, now  
 abandoned  
 DT Utility  
 FS Granted  
 EXNAM Primary Examiner: Evans, J. E.  
 LREP Shipley, James E.  
 CLMN Number of Claims: 13  
 ECL Exemplary Claim: 1  
 DRWN No Drawings



10/631,862

LN.CNT 397

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB An improved process for the manufacture of 1,1,1,2-tetrafluoroethane, more particularly, a gas-phase reaction of a 1,1,1-trifluorochloroethane with hydrogen fluoride in the presence of a selected metal on aluminum fluoride or carbon.

L2 ANSWER 82 OF 82 JAPIO (C) 2004 JPO on STN

AN 2004-068007 JAPIO

TI METHOD FOR PRODUCING PERFLUOROPOLYETHERS HAVING ALDEHYDE, ALCOHOL, AND AMINE TERMINAL GROUPS

IN DI MEO ANTONELLO; PICOZZI ROSALDO; TONELLI CLAUDIO

PA SOLVAY SOLEXIS SPA

PI JP 2004068007 A 20040304 Heisei

AI JP 2003-205414 (JP2003205414 Heisei) 20030801

PRAI IT 2002-MI02 1734 20020801

SO PATENT ABSTRACTS OF JAPAN (CD-ROM), Unexamined Applications, Vol. 2004

AB PROBLEM TO BE SOLVED: To provide a method for producing a reduced compound having a corresponding aldehyde, alcohol, or amine terminal group in a high yield of  $\geq 90\%$  from precursors of perfluoropolyethers having an acyl-chloride, aldehyde or nitrile terminal group.

SOLUTION: The problem is solved by the method for producing a perfluoropolyether having a reactive terminal group of  $-\text{CH}(\text{SB})_2\text{NH}(\text{SB})_2$ ,  $-\text{CHO}$ , or  $\text{CH}(\text{SB})_2\text{OH}$  by reducing the corresponding perfluoropolyether having a  $-\text{CN}$ ,  $-\text{COCl}$ , or  $-\text{CHO}$  terminal group using a hydrogen gas in the presence of a catalyst constituted with Pd, Rh, and Ru supported on a solid metallic fluoride at a temperature of  $20\text{--}150^\circ\text{C}$  under a pressure of 1-50 atmospheric  
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Patent Office Classifications  
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(Version 7.01 for Windows) now available  
NEWS 7 AUG 27 BIOCOMMERCE: Changes and enhancements to content coverage  
NEWS 8 AUG 27 BIOTECHABS/BIOTECHDS: Two new display fields added for legal  
status data from INPADOC  
NEWS 9 SEP 01 INPADOC: New family current-awareness alert (SDI) available  
NEWS 10 SEP 01 New pricing for the Save Answers for SciFinder Wizard within  
STN Express with Discover!  
NEWS 11 SEP 01 New display format, HITSTR, available in WPIDS/WPINDEX/WPIX  
NEWS 12 SEP 27 STANDARDS will no longer be available on STN  
NEWS 13 SEP 27 SWETSCAN will no longer be available on STN  
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10/631,862

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DICTIONARY FILE UPDATES: 8 NOV 2004 HIGHEST RN 777024-10-9

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G1 O CFCF2O1-10 F O0-10 F O1-10 F O0-10 F G12 22 26 27 28 29 30 31 34 35 36 37 40 41 45  
F 23 32 38 42

F<sup>1</sup>  
2F<sup>2</sup>CF<sub>3</sub> 5CH<sub>2</sub>OH  
3CF<sub>3</sub>CF<sub>2</sub>CF<sub>3</sub> 6CH<sub>2</sub>NH<sub>2</sub>  
Cl<sub>3</sub><sup>4</sup> C<sub>6</sub>H<sub>7</sub>

1<sup>1</sup>  
2<sup>2</sup>7 5<sup>3</sup>10  
3<sup>3</sup>46 6<sup>9</sup>11  
5<sup>4</sup> 12<sup>7</sup>

chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 21 22 23 25 26 27 28 29 30 31 32  
33 34 35 36 37 38 39 40 41 42 43 45

chain bonds :

2-7 3-4 4-6 8-10 9-11 21-22 22-23 22-25 22-26 26-27 27-28 28-29 29-30  
30-31 31-32 31-33 31-34 34-35 35-36 36-37 37-40 37-38 37-39 40-41 41-42  
41-43 41-45

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exact/norm bonds :

21-22 22-25 22-26 31-34 36-37 37-40 40-41 41-43 41-45

exact bonds :

2-7 3-4 4-6 8-10 9-11 22-23 26-27 27-28 28-29 29-30 30-31 31-32 31-33  
34-35 35-36 37-38 37-39 41-42

G1: [\*1], [\*2], [\*3], [\*4], [\*5], [\*6], [\*7]

G2: [\*1], [\*2]

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS  
10:CLASS 11:CLASS 12:CLASS 21:CLASS 22:CLASS 23:CLASS 25:CLASS 26:CLASS  
27:CLASS 28:CLASS 29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS  
35:CLASS 36:CLASS 37:CLASS 38:CLASS 39:CLASS 40:CLASS 41:CLASS 42:CLASS  
43:CLASS 45:CLASS

L2 STRUCTURE UPLOADED

=> que L2 NOT L1

L3 QUE L2 NOT L1

=> s l3

STRUCTURE TOO LARGE - SEARCH ENDED

A structure in your query is too large. You may delete attributes or atoms to reduce the size of the structure and try again.

=> file stnguide

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	0.84	1.26

FILE 'STNGUIDE' ENTERED AT 06:58:40 ON 10 NOV 2004  
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FILE CONTAINS CURRENT INFORMATION.  
LAST RELOADED: Nov 5, 2004 (20041105/UP).

=> file reg

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	0.36	1.62

FILE 'REGISTRY' ENTERED AT 07:02:23 ON 10 NOV 2004  
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STRUCTURE FILE UPDATES: 8 NOV 2004 HIGHEST RN 777024-10-9  
DICTIONARY FILE UPDATES: 8 NOV 2004 HIGHEST RN 777024-10-9

10/631,862

TSCA INFORMATION NOW CURRENT THROUGH MAY 21, 2004

Please note that search-term pricing does apply when  
conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. For more  
information enter HELP PROP at an arrow prompt in the file or refer  
to the file summary sheet on the web at:  
<http://www.cas.org/ONLINE/DBSS/registryss.html>

=> ....Testing the current file.... screen

ENTER SCREEN EXPRESSION OR (END):end

=> screen 1994 OR 2016 OR 2021 OR 2026 OR 1929 OR 1838

L4 SCREEN CREATED

=>

Uploading C:\Program Files\Stnexp\Queries\10630698a.str

G2 CF3 CF3 G2 25 33 39 43  
G1 O CFCF2O1-5 F2 O O-5 CFCF1-5 F O O-5 G1 12 26 27 28 29 30 31 34 35 36 37 40 41 45  
F F F F 23 32 38 42



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 21 22 23 25 26 27 28 29 30 31 32  
33 34 35 36 37 38 39 40 41 42 43 45

chain bonds :

2-7 3-4 4-6 8-10 9-11 21-22 22-23 22-25 22-26 26-27 27-28 28-29 29-30  
30-31 31-32 31-33 31-34 34-35 35-36 36-37 37-40 37-38 37-39 40-41 41-42  
41-43 41-45

exact/norm bonds :

21-22 22-25 22-26 31-34 36-37 37-40 40-41 41-43 41-45

exact bonds :

2-7 3-4 4-6 8-10 9-11 22-23 26-27 27-28 28-29 29-30 30-31 31-32 31-33  
34-35 35-36 37-38 37-39 41-42

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G1:[\*1],[\*2],[\*3],[\*4],[\*5],[\*6],[\*7]

G2:[\*1],[\*2]

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS  
10:CLASS 11:CLASS 12:CLASS 21:CLASS 22:CLASS 23:CLASS 25:CLASS 26:CLASS  
27:CLASS 28:CLASS 29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS  
35:CLASS 36:CLASS 37:CLASS 38:CLASS 39:CLASS 40:CLASS 41:CLASS 42:CLASS  
43:CLASS 45:CLASS

L5 STRUCTURE UPLOADED

=> que L5 NOT L4

L6 QUE L5 NOT L4

=> s l6

STRUCTURE TOO LARGE - SEARCH ENDED

A structure in your query is too large. You may delete attributes or atoms to reduce the size of the structure and try again.

=> file stnguide

COST IN U.S. DOLLARS

SINCE FILE

TOTAL

ENTRY

SESSION

FULL ESTIMATED COST

0.84

2.46

FILE 'STNGUIDE' ENTERED AT 07:03:35 ON 10 NOV 2004

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FILE CONTAINS CURRENT INFORMATION.

LAST RELOADED: Nov 5, 2004 (20041105/UP).

=> file reg

COST IN U.S. DOLLARS

SINCE FILE

TOTAL

ENTRY

SESSION

FULL ESTIMATED COST

0.30

2.76

FILE 'REGISTRY' ENTERED AT 07:06:51 ON 10 NOV 2004

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STRUCTURE FILE UPDATES: 8 NOV 2004 HIGHEST RN 777024-10-9

DICTIONARY FILE UPDATES: 8 NOV 2004 HIGHEST RN 777024-10-9

TSCA INFORMATION NOW CURRENT THROUGH MAY 21, 2004

Please note that search-term pricing does apply when conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details.

10/631,862

Experimental and calculated property data are now available. For more information enter HELP PROP at an arrow prompt in the file or refer to the file summary sheet on the web at:  
<http://www.cas.org/ONLINE/DBSS/registryss.html>

=> ....Testing the current file.... screen

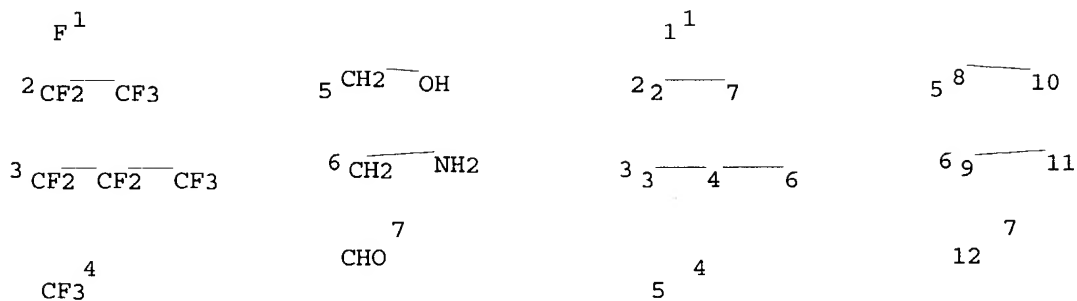
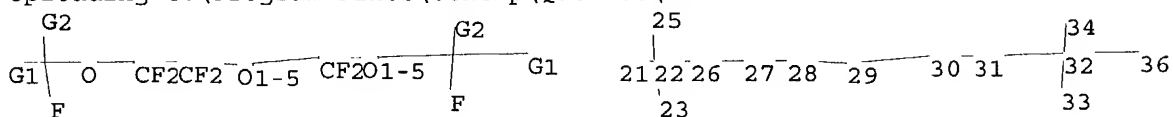
ENTER SCREEN EXPRESSION OR (END):end

=> screen 1994 OR 2016 OR 2021 OR 2026 OR 1929 OR 1838

L7 SCREEN CREATED

=>

Uploading C:\Program Files\Stnexp\Queries\10630698b.str



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 21 22 23 25 26 27 28 29 30 31 32  
33 34 36

chain bonds :

2-7 3-4 4-6 8-10 9-11 21-22 22-23 22-25 22-26 26-27 27-28 28-29 29-30  
30-31 31-32 32-36 32-33 32-34

exact/norm bonds :

21-22 22-25 22-26 31-32 32-36 32-34

exact bonds :

2-7 3-4 4-6 8-10 9-11 22-23 26-27 27-28 28-29 29-30 30-31 32-33

10/631,862

G1:[\*1],[\*2],[\*3],[\*4],[\*5],[\*6],[\*7]

G2:[\*1],[\*2]

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS  
10:CLASS 11:CLASS 12:CLASS 21:CLASS 22:CLASS 23:CLASS 25:CLASS 26:CLASS  
27:CLASS 28:CLASS 29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS  
36:CLASS

L8 STRUCTURE UPLOADED

=> que L8 NOT L7

L9 QUE L8 NOT L7

=> d

L9 HAS NO ANSWERS

L7 SCR 1994 OR 2016 OR 2021 OR 2026 OR 1929 OR 1838

L8 STR

\* STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT \*

Structure attributes must be viewed using STN Express query preparation.

L9 QUE L8 NOT L7

=> s l9

SAMPLE SEARCH INITIATED 07:07:31 FILE 'REGISTRY'

SAMPLE SCREEN SEARCH COMPLETED - 0 TO ITERATE

100.0% PROCESSED 0 ITERATIONS 0 ANSWERS  
SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE \*\*COMPLETE\*\*  
BATCH \*\*COMPLETE\*\*

PROJECTED ITERATIONS: 0 TO 0

PROJECTED ANSWERS: 0 TO 0

L10 0 SEA SSS SAM L8 NOT L7

=> s l9 ful

FULL SEARCH INITIATED 07:07:40 FILE 'REGISTRY'

FULL SCREEN SEARCH COMPLETED - 0 TO ITERATE

100.0% PROCESSED 0 ITERATIONS 0 ANSWERS  
SEARCH TIME: 00.00.01

L11 0 SEA SSS FUL L8 NOT L7

=> file stnguide

COST IN U.S. DOLLARS

SINCE FILE

TOTAL

ENTRY

SESSION

FULL ESTIMATED COST

156.26

159.02

FILE 'STNGUIDE' ENTERED AT 07:08:21 ON 10 NOV 2004

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FILE CONTAINS CURRENT INFORMATION.

LAST RELOADED: Nov 5, 2004 (20041105/UP).

=> file reg

COST IN U.S. DOLLARS

SINCE FILE

TOTAL

ENTRY

SESSION

FULL ESTIMATED COST

0.54

159.56

FILE 'REGISTRY' ENTERED AT 07:13:48 ON 10 NOV 2004

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STRUCTURE FILE UPDATES: 8 NOV 2004 HIGHEST RN 777024-10-9

DICTIONARY FILE UPDATES: 8 NOV 2004 HIGHEST RN 777024-10-9

TSCA INFORMATION NOW CURRENT THROUGH MAY 21, 2004

Please note that search-term pricing does apply when  
conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. For more  
information enter HELP PROP at an arrow prompt in the file or refer  
to the file summary sheet on the web at:  
<http://www.cas.org/ONLINE/DBSS/registryss.html>

=> ....Testing the current file.... screen

ENTER SCREEN EXPRESSION OR (END):end

=> screen 1994 OR 2016 OR 2021 OR 2026 OR 1929 OR 1838

L12 SCREEN CREATED

=>

Uploading C:\Program Files\Stnexp\Queries\10630698c.str



10/631,862

=> s l14

STRUCTURE TOO LARGE - SEARCH ENDED

A structure in your query is too large. You may delete attributes or atoms to reduce the size of the structure and try again.

=> file stnguide

COST IN U.S. DOLLARS

SINCE FILE	TOTAL
ENTRY	SESSION
0.84	160.40

FULL ESTIMATED COST

FILE 'STNGUIDE' ENTERED AT 07:14:45 ON 10 NOV 2004  
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FILE CONTAINS CURRENT INFORMATION.

LAST RELOADED: Nov 5, 2004 (20041105/UP).

=> file reg

COST IN U.S. DOLLARS

SINCE FILE	TOTAL
ENTRY	SESSION
0.30	160.70

FULL ESTIMATED COST

FILE 'REGISTRY' ENTERED AT 07:17:49 ON 10 NOV 2004  
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STRUCTURE FILE UPDATES: 8 NOV 2004 HIGHEST RN 777024-10-9  
DICTIONARY FILE UPDATES: 8 NOV 2004 HIGHEST RN 777024-10-9

TSCA INFORMATION NOW CURRENT THROUGH MAY 21, 2004

Please note that search-term pricing does apply when conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. For more information enter HELP PROP at an arrow prompt in the file or refer to the file summary sheet on the web at:  
<http://www.cas.org/ONLINE/DBSS/registryss.html>

=> ....Testing the current file.... screen

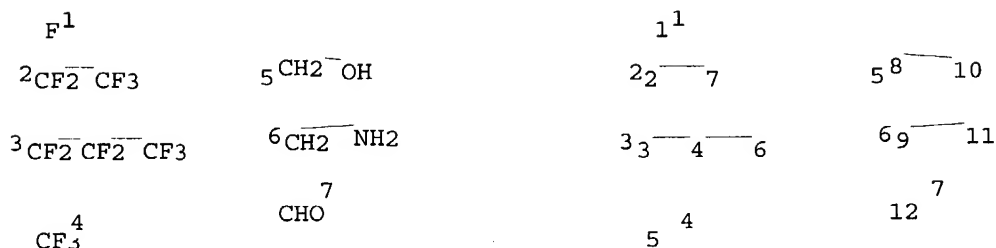
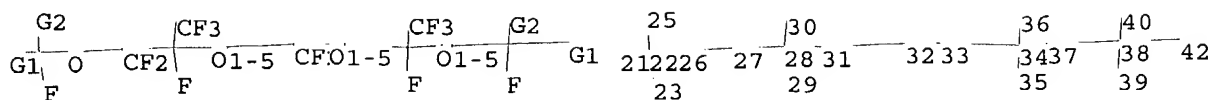
ENTER SCREEN EXPRESSION OR (END):end

=> screen 1994 OR 2016 OR 2021 OR 2026 OR 1838

L15 SCREEN CREATED

=>  
Uploading C:\Program Files\Stnexp\Queries\10630698d.str

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chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 21 22 23 25 26 27 28 29 30 31 32  
33 34 35 36 37 38 39 40 42

chain bonds :

2-7 3-4 4-6 8-10 9-11 21-22 22-23 22-25 22-26 26-27 27-28 28-29 28-30  
28-31 31-32 32-33 33-34 34-37 34-35 34-36 37-38 38-39 38-40 38-42

exact/norm bonds :

21-22 22-25 22-26 28-31 33-34 34-37 37-38 38-40 38-42

exact bonds :

2-7 3-4 4-6 8-10 9-11 22-23 26-27 27-28 28-29 28-30 31-32 32-33 34-35  
34-36 38-39

G1: [\*1], [\*2], [\*3], [\*4], [\*5], [\*6], [\*7]

G2: [\*1], [\*2]

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS  
10:CLASS 11:CLASS 12:CLASS 21:CLASS 22:CLASS 23:CLASS 25:CLASS 26:CLASS  
27:CLASS 28:CLASS 29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS  
35:CLASS 36:CLASS 37:CLASS 38:CLASS 39:CLASS 40:CLASS 42:CLASS

L16 STRUCTURE UPLOADED

=> que L16 NOT L15

L17 QUE L16 NOT L15

10/631,862

=> s 117

SAMPLE SEARCH INITIATED 07:18:20 FILE 'REGISTRY'  
SAMPLE SCREEN SEARCH COMPLETED - 174 TO ITERATE

100.0% PROCESSED 174 ITERATIONS ( 10 INCOMPLETE) 10 ANSWERS  
SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE \*\*COMPLETE\*\*  
BATCH \*\*COMPLETE\*\*  
PROJECTED ITERATIONS: 2689 TO 4271  
PROJECTED ANSWERS: 11 TO 389

L18 10 SEA SSS SAM L16 NOT L15

=> s 117 ful

FULL SEARCH INITIATED 07:18:29 FILE 'REGISTRY'  
FULL SCREEN SEARCH COMPLETED - 3484 TO ITERATE

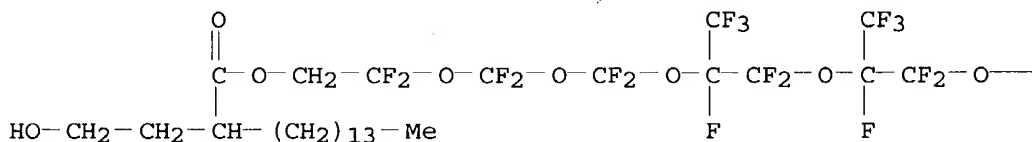
100.0% PROCESSED 3484 ITERATIONS ( 156 INCOMPLETE) 156 ANSWERS  
SEARCH TIME: 00.00.07

L19 156 SEA SSS FUL L16 NOT L15

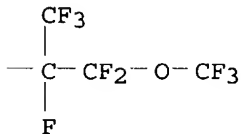
=> d scan

L19 156 ANSWERS REGISTRY COPYRIGHT 2004 ACS on STN  
ITERATION INCOMPLETE  
IN Hexadecanoic acid, 2-(2-hydroxyethyl)-, 2,2,4,4,6,6,8,9,9,11,12,12,14,15,15,17,17,17-octadecafluoro-8,11,14-tris(trifluoromethyl)-3,5,7,10,13,16-hexaoxaheptadec-1-yl ester (9CI)  
MF C32 H37 F27 O9

PAGE 1-A



PAGE 1-B



\*\*PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT\*\*

HOW MANY MORE ANSWERS DO YOU WISH TO SCAN? (1):file caplus casreact uspatful  
'FILE CAPLUS CASREACT USPATFUL' IS NOT VALID HERE

To display more answers, enter the number of answers you would like to see. To end the display, enter "NONE", "N", "0", or "END".

HOW MANY MORE ANSWERS DO YOU WISH TO SCAN? (1):0

10/631,862

=> file caplus casreact uspatful  
COST IN U.S. DOLLARS

SINCE FILE	TOTAL
ENTRY	SESSION
156.68	317.38

FULL ESTIMATED COST

FILE 'CAPLUS' ENTERED AT 07:20:01 ON 10 NOV 2004  
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FILE 'CASREACT' ENTERED AT 07:20:01 ON 10 NOV 2004  
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FILE 'USPATFULL' ENTERED AT 07:20:01 ON 10 NOV 2004  
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=> s l19

L20 120 L19

=> dup rem l20

PROCESSING COMPLETED FOR L20

L21 108 DUP REM L20 (12 DUPLICATES REMOVED)

=> d 1-108 ti

L21 ANSWER 1 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Manufacture of magnetic recording media

L21 ANSWER 2 OF 108 USPATFULL on STN  
TI Novel aerosol formulation containing a polar fluorinated molecule

L21 ANSWER 3 OF 108 USPATFULL on STN  
TI Amphiphilic lubricants for magnetic media

L21 ANSWER 4 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 1  
TI Amphiphilic lubricants with central polar polyfunctional group and pair of fluoroalkylether endgroups as topcoats for magnetic recording media

L21 ANSWER 5 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Aerosol formulation containing a polar fluorinated compound

L21 ANSWER 6 OF 108 USPATFULL on STN  
TI Process for thermal decomposition of hexafluoropropylene oxide oligomers

L21 ANSWER 7 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 2  
TI Perfluorinated organo substituted cyclosiloxanes and copolymers prepared from these cyclosiloxanes

L21 ANSWER 8 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 3  
TI Perfluorinated ether organo substituted cyclosiloxanes and siloxane (co)polymers prepared from these cyclosiloxanes

L21 ANSWER 9 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Magnetic recording medium with fluorine-containing alkylcarboxylic acid lubricating layer

L21 ANSWER 10 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Process for thermal decomposition of hexafluoropropylene oxide oligomers

L21 ANSWER 11 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN

10/631,862

- TI Process for preparation of hexafluoropropylene oxide by oxidation of hexafluoropropylene
- L21 ANSWER 12 OF 108 USPATFULL on STN  
TI Fluoroalkylated amphiphilic ligands, their metallic complexes and their uses
- L21 ANSWER 13 OF 108 USPATFULL on STN  
TI Amides and esters of perfluoropolyoxaalkylene-sulfo- or perfluoropolyoxaalkylene-carboxylic acids and a process for producing same
- L21 ANSWER 14 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Water Core within Perfluoropolyether-Based Microemulsions Formed in Supercritical Carbon Dioxide
- L21 ANSWER 15 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Adsorption of fluorine-containing surfactants from aqueous solutions on the surface of polyamide fibers
- L21 ANSWER 16 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Radical additions to fluoroolefins. Photochemical mono-fluoroalkylation and sequential bis-fluoroalkylation of oxolane
- L21 ANSWER 17 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Effect of colloidal-chemical properties of fluorine-containing latexes and fluorocarbon surfactants on the modification of textiles
- L21 ANSWER 18 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 4  
TI Fluorine-containing alkylsuccinic acid diester and its preparation and use as a lubricant for magnetic recording media
- L21 ANSWER 19 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Manufacture of amides and esters of perfluoropolyoxyalkylenesulfonic or -carboxylic acids
- L21 ANSWER 20 OF 108 USPATFULL on STN  
TI Fluoroalkylated amphiphilic ligands and their metallic complexes
- L21 ANSWER 21 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Reaction of perfluoropolyoxapolypropenecarboxylic acids with metal carbonates and acid fluorides with 3-Amino-1,2,4-Triazole
- L21 ANSWER 22 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Esterification of perfluoropolyoxapolypropylenecarboxylic acid ( $n = 8$ )
- L21 ANSWER 23 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Fluoroalkylated amphiphilic ligands, their metallic complexes and their uses
- L21 ANSWER 24 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Preparation of fluoroalkanoic acid esters and magnetic recording medium with lubricant layer containing them
- L21 ANSWER 25 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Preparation of fluorinated alcohols and magnetic recording media using them as lubricants
- L21 ANSWER 26 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Perfluoroalkyl group-containing polymers
- L21 ANSWER 27 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN

10/631,862

- TI Perfluorooxalkyl group-terminated vinyl polymers
- L21 ANSWER 28 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Preparation of polyfluoroalkanoyl peroxides as polymerization initiators
- L21 ANSWER 29 OF 108 USPATFULL on STN  
TI Polyfluoroalkanoyl peroxide
- L21 ANSWER 30 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Surface activity of fluorine-containing surfactants in polar solvents and water-organic mixtures
- L21 ANSWER 31 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Preparation of perfluoroacetal and perfluoroketal compounds and use thereof in thermal shock testing
- L21 ANSWER 32 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Methylcarbinol-terminated hexafluoropropylene oxide oligoether derivatives
- L21 ANSWER 33 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Preparation of carbonyl fluorides by oligomerization of hexafluoropropene oxides
- L21 ANSWER 34 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Process and catalysts for the manufacture of hexafluoropropylene oxide oligomers
- L21 ANSWER 35 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Electrolytic decarboxylation of perfluorocarboxylic acids or their soluble salts and subsequent dimerization of the radicals produced
- L21 ANSWER 36 OF 108 USPATFULL on STN  
TI Process for the oligomerization of hexafluoropropene oxide
- L21 ANSWER 37 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Aggregation of perfluorinated polymers in aqueous solution studied by ESR
- L21 ANSWER 38 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Process for preparation of perfluorinated carboxylic acid fluorides
- L21 ANSWER 39 OF 108 USPATFULL on STN  
TI Process for the preparation of perfluorinated carbonyl fluorides
- L21 ANSWER 40 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 5  
TI Preparation of carbonyl fluoride compounds
- L21 ANSWER 41 OF 108 USPATFULL on STN  
TI Fluoropolyethers containing end groups endowed with anchoring capacity
- L21 ANSWER 42 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Fuel cells
- L21 ANSWER 43 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 6  
TI Fluorine-containing methacrylate esters
- L21 ANSWER 44 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Fluorine-containing polymers with oxygen permeability for medical use
- L21 ANSWER 45 OF 108 USPATFULL on STN  
TI Shaped article of synthetic resin having improved surface
- L21 ANSWER 46 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN



- TI Synthetic resin films with water and oil repellence
- L21 ANSWER 47 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Lubricant finishes
- L21 ANSWER 48 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Fluoropolyether compounds
- L21 ANSWER 49 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Surface active substances containing an oligo(hexafluoropropene oxide) chain as a hydrophobic oleophobic moiety
- L21 ANSWER 50 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Acrylic acid esters
- L21 ANSWER 51 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Water and oil repellents
- L21 ANSWER 52 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Synthesis of fluorinated surfactants containing hexafluoropropene oxide as a hydrophobic group and properties of the solutions
- L21 ANSWER 53 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Polymerization of fluorine-containing monomers
- L21 ANSWER 54 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Foaming agent for extinguishing fires
- L21 ANSWER 55 OF 108 USPATFULL on STN
- TI Alkyl perfluoro- $\omega$ -fluoroformyl esters and their preparation
- L21 ANSWER 56 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Methods of calculating engineering parameters for gas separations
- L21 ANSWER 57 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 7
- TI Alkyl perfluoro- $\omega$ -fluoroformyl esters and monomers therefrom
- L21 ANSWER 58 OF 108 USPATFULL on STN
- TI Process for the preparation of fluorine-containing ketones
- L21 ANSWER 59 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Alkylperfluoro- $\omega$ -fluoroformyl esters
- L21 ANSWER 60 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Perfluoro ketones
- L21 ANSWER 61 OF 108 USPATFULL on STN
- TI Alkyl perfluoro- $\omega$ -fluoroformyl esters and their preparation
- L21 ANSWER 62 OF 108 USPATFULL on STN
- TI Fluorocarbon triazine polymers
- L21 ANSWER 63 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Synthesis of perfluoro(polyether) difunctional compounds
- L21 ANSWER 64 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN
- TI Methods for the estimation of vapor pressures and oxygen solubilities of fluorochemicals for possible application in artificial blood formulations
- L21 ANSWER 65 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 8
- TI Fluorocarbon dye dispersion for exhaust disperse dyeing

10/631,862

- L21 ANSWER 66 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Fluorocarbon triazine polymers
- L21 ANSWER 67 OF 108 USPATFULL on STN  
TI Fluoroalkyleneether difunctional compounds
- L21 ANSWER 68 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Study of the kinetics of the reaction of hexafluoropropylene oxide with organic salts in a medium of aprotic solvents
- L21 ANSWER 69 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Oligomeric fluorinated additives as surface modifiers for solid polymers
- L21 ANSWER 70 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI The solubility of oxygen in highly fluorinated liquids
- L21 ANSWER 71 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 9  
TI Rapid fixation of disperse dyes on synthetic polymers
- L21 ANSWER 72 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 10  
TI Displacement of organic liquid films from solid surfaces by nonaqueous systems
- L21 ANSWER 73 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Exhaust dyeing of synthetic polymers with dyes dispersed in solution or emulsion in a saturated liquid fluorocarbon
- L21 ANSWER 74 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Perfluorinated ethers
- L21 ANSWER 75 OF 108 USPATFULL on STN  
TI Process for preparing perfluorinated ethers
- L21 ANSWER 76 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Colloidal-chemical properties of solutions of surfactants based on perfluoropropylene oxide oligomers. 1. Surface activity of ammonium salts of perfluorooligoestermonocarboxylic acids at the aqueous solution-air interface
- L21 ANSWER 77 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Oligomeric fluorinated additives as surface modifiers for solid polymers
- L21 ANSWER 78 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 11  
TI Exhaust disperse dyeing of synthetic polymers using a saturated liquid fluorocarbon
- L21 ANSWER 79 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 12  
TI Bis-triazine compounds
- L21 ANSWER 80 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Exhaust disperse dyeing of synthetic fibers
- L21 ANSWER 81 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Fluoroalkylene ether difunctional compounds
- L21 ANSWER 82 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Dyeing synthetic fabrics with disperse dyes in fluorocarbon solvents
- L21 ANSWER 83 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Dispersion of dye in a fluorocarbon for exhaust dyeing
- L21 ANSWER 84 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN

- TI Rapid fixation of disperse dyes on synthetic polymers
- L21 ANSWER 85 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Fluorocarbon-dye dispersion for exhaust dispersion dyeing
- L21 ANSWER 86 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Exhaustion dyeing of films, fibers, and textiles of synthetic polymers with disperse dyes
- L21 ANSWER 87 OF 108 USPATFULL on STN  
 TI Acrylic and methacrylic monomers, polymers and copolymers
- L21 ANSWER 88 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Perfluorinated linear polyethers having reactive terminal groups at both ends of the chain
- L21 ANSWER 89 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Perfluoroalkyletheramidoalkyl betaines and sulfobetaines
- L21 ANSWER 90 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI  $\alpha,\omega$ -Di-s-triazinyl perfluorooxaalkanes
- L21 ANSWER 91 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Polyfluoroalkoxy alkyl amidocarboxylic acids and salts
- L21 ANSWER 92 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Solid lubricant additives dispersed in perfluoroalkyl ethers with perfluoroalkyl ether acid dispersants
- L21 ANSWER 93 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Perfluoropolyethers by photooxidation of fluoroolefins
- L21 ANSWER 94 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Perfluoropoly(ether esters) as lubricants and hydraulic fluids
- L21 ANSWER 95 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Esters of hexafluoropropylene oxide polymer acids and polyalkylene glycols
- L21 ANSWER 96 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Liquid phase decarbonylation of fluorinated acyl fluorides
- L21 ANSWER 97 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Acrylate-type esters of perfluoropolyoxaalkaneamidoalkyl alcohols, and their polymers which are useful as oil and water repellents and as metal corrosion inhibitors
- L21 ANSWER 98 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Elastomers comprising acrylic and methacrylic derivatives of polyfluoropolyethers
- L21 ANSWER 99 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Polyfluoropolyoxaalkyl acrylates and N-(polyfluoropolyoxaalkyl)acrylamides
- L21 ANSWER 100 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Perfluoroalkyl ether amidoamine oxides
- L21 ANSWER 101 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Oil repellent polyfluoropolyoxo-alkyl phosphates
- L21 ANSWER 102 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Vaporization and decomposition kinetics of candidate re-entry blackout suppressants in low-pressure flames

L21 ANSWER 103 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI High temperature study of electrophilic gases for plasma quenching

L21 ANSWER 104 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Corrosion-inhibited poly(hexafluoropropylene oxide) lubricants

L21 ANSWER 105 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Perfluoro ketones

L21 ANSWER 106 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Perfluorinated polyethers. Synthesis and characterization of a new class of inert fluids

L21 ANSWER 107 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Fluorocarbon ethers from hexafluoropropylene oxide

L21 ANSWER 108 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 TI Dicarboxylic acids of fluorocarbon ethers and fluorides and their esters, amides, and salts

=> d 38,39,40,41,44,48,49,52,63,67,74,75,81,88,89,100,106 bib ab fhitr

L21 ANSWER 38 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 AN 1989:614127 CAPLUS  
 DN 111:214127  
 TI Process for preparation of perfluorinated carboxylic acid fluorides  
 IN Kruse, Alfred; Siegemund, Guenter; Schwertfeger, Werner  
 PA Hoechst A.-G., Fed. Rep. Ger.  
 SO Ger. Offen., 5 pp.  
 CODEN: GWXXBX  
 DT Patent  
 LA German  
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	DE 3737920	A1	19890518	DE 1987-3737920	19871107
	US 4874557	A	19891017	US 1988-266919	19881103
	EP 315908	A1	19890517	EP 1988-118391	19881104
	EP 315908	B1	19920826		
	R: BE, CH, DE, FR, GB, IT, LI, NL				
	JP 01157933	A2	19890621	JP 1988-277536	19881104
	CN 1034199	A	19890726	CN 1988-107738	19881107
	CN 1022240	B	19930929		
PRAI	DE 1987-3737920		19871107		

OS MARPAT 111:214127  
 AB F3CCF2[CF2OCF(CF3)]<sub>n</sub>COF (I; n = 2, 3), useful intermediates and monomers, are prepared by oligomerization of hexafluoropropylene oxide (II) at -20 to +100° in the presence of a catalyst system comprising: 1) alkali fluoride, preferably KF, 2-30%; 2) C5-8 alkanedinitrile, preferably adiponitrile, 50-95%; 3) MeO(CH<sub>2</sub>CH<sub>2</sub>O)<sub>m</sub>Me (III; m = 2-6, preferably 3) 2-50%. The process is advantageous in that higher temps. are used, product composition can be controlled by manipulation of the catalyst system composition, the product is readily separated, and the catalyst system can be reused. Thus, in a stainless steel autoclave a mixture of 30 g KF, 500 mL adiponitrile, and 100 mL III (m = 3) was stirred 30 min., continuously pressurized to 3.5 bar by addition of 5 kg II and stirred 2.5 h at 35-40°. After 3 h addnl. stirring the mixture readily separated into 2 phases. The lower product phase (4.90 kg) was drawn off and comprised the following I: n = 1, 21.5; n = 2, 61.1; n = 3, 16.3; and n = 4, 0.8%.

IT 13252-15-8P

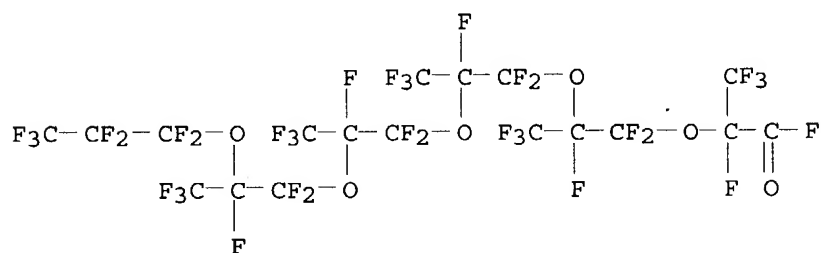
10/631,862

RL: PREP (Preparation)

(manufacture of, by oligomerization of hexafluoropropylene oxide, catalysts for)

RN 13252-15-8 CAPLUS

CN 3,6,9,12,15-Pentaoxaoctadecanoyl fluoride, 2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,17,17,18,18,18-eicosafuoro-2,5,8,11,14-pentakis(trifluoromethyl)-(7CI, 8CI, 9CI) (CA INDEX NAME)



L21 ANSWER 39 OF 108 USPATFULL on STN

AN 89:85709 USPATFULL

TI Process for the preparation of perfluorinated carbonyl fluorides

IN Kruse, Alfred, Kelkheim, Germany, Federal Republic of  
Siegemund, Gunter, Hofheim am Taunus, Germany, Federal Republic of  
Schwertfeger, Werner, Langgons, Germany, Federal Republic of

PA Hoechst Aktiengesellschaft, Frankfurt am Main, Germany, Federal Republic  
of (non-U.S. corporation)

PI US 4874557 19891017

AI US 1988-266919 19881103 (7)

PRAI DE 1987-3737920 19871107

DT Utility

FS Granted

EXNAM Primary Examiner: Killos, Paul J.

CLMN Number of Claims: 8

ECL Exemplary Claim: 1

DRWN No Drawings

LN.CNT 238

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

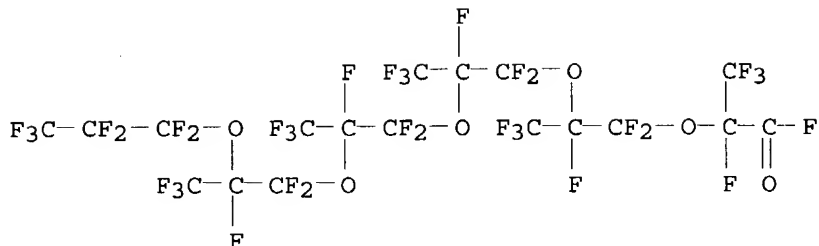
AB The invention relates to a process for the preparation of perfluorinated carbonyl fluorides of the formula ##STR1## by oligomerization of hexafluoropropene oxide in the presence of a catalyst. The catalyst comprises a mixture of an alkali metal fluoride, a carboxylic acid dinitrile and a polyethylene glycol dimethyl ether.

IT 13252-15-8P

(manufacture of, by oligomerization of hexafluoropropylene oxide, catalysts for)

RN 13252-15-8 USPATFULL

CN 3,6,9,12,15-Pentaoxaoctadecanoyl fluoride, 2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,17,17,18,18,18-eicosafuoro-2,5,8,11,14-pentakis(trifluoromethyl)-(7CI, 8CI, 9CI) (CA INDEX NAME)



L21 ANSWER 40 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 5

AN 1989:438876 CAPLUS

DN 111:38876

TI Preparation of carbonyl fluoride compounds

IN Okabe, Jun; Tatsu, Haruyoshi

PA Nippon Mectron Co., Ltd., Japan

SO U.S., 7 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 4769184	A	19880906	US 1987-121135	19871116
	JP 01066139	A2	19890313	JP 1987-222946	19870908
	JP 08019035	B4	19960228		
	JP 01093557	A2	19890412	JP 1987-249588	19871002
	JP 2726824	B2	19980311		
PRAI	JP 1987-222946		19870908		
	JP 1987-249588		19871002		

OS MARPAT 111:38876

AB A process for producing XCOF (I; X = F, CF<sub>3</sub>) or I (X = CF<sub>3</sub>CF<sub>2</sub>), useful as intermediates for producing perfluoro(alkyl vinyl ethers) which are monomers for producing F-containing resins, F-containing rubber, etc., comprised

thermally decomposing RfO(CF<sub>2</sub>CF<sub>2</sub>O)<sub>a</sub>(CF<sub>2</sub>O)<sub>b</sub>(O)cRf' (Rf = perfluoroalkyl; Rf' = COF, CF<sub>3</sub>; the CF<sub>2</sub>O and O groups are distributed at random; a, b ≠ 0; c can be 0; a + b + c ≤ .apprx.200) or RfO(CFXCF<sub>2</sub>O)<sub>n</sub>CFX'Y (X' = CF<sub>3</sub>, F, H; Y = COF, CO<sub>2</sub>H, CO<sub>2</sub>R, CF<sub>3</sub>; R = alkyl; n = 1-50), resp. F<sub>2</sub>C:CF<sub>2</sub> and O<sub>2</sub> were irradiated with UV to give F<sub>3</sub>CO(CF<sub>2</sub>OF<sub>2</sub>O)<sub>8</sub>(CF<sub>2</sub>O)<sub>24</sub>O<sub>0.4</sub>COF, thermal decomposition of which at 200° over activated C gave a mixture of 78.2% COF<sub>2</sub> and 21.8% F<sub>3</sub>CCOF. I (X = F, CF<sub>3</sub>) so produced contain no Cl-based impurities.

IT 13140-28-8P

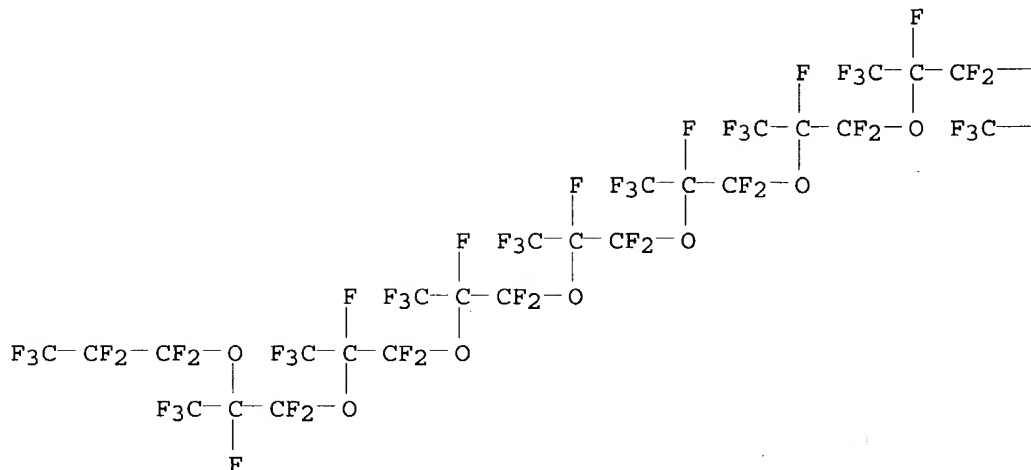
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and reaction of, in synthesis of carbonyl fluorides)

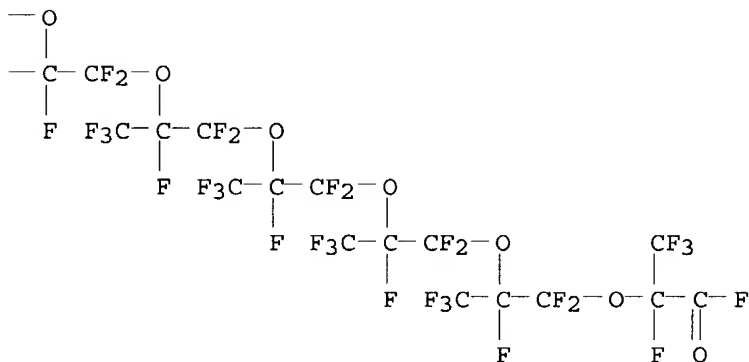
RN 13140-28-8 CAPLUS

CN 3,6,9,12,15,18,21,24,27,30,33,36,39-Tridecaoxadotetracontanoyl fluoride, 2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,17,19,19,20,22,22,23,25,25,26,28,28, 29,31,31,32,34,34,35,37,37,38,40,40,41,41,42,42,42-tetratetracontafluoro-2,5,8,11,14,17,20,23,26,29,32,35,38-tridecakis(trifluoromethyl)- (7CI, 8CI, 9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 1-B



L21 ANSWER 41 OF 108 USPATFULL on STN  
 AN 88:5708 USPATFULL  
 TI Fluoropolyethers containing end groups endowed with anchoring capacity  
 IN Caporiccio, Gerardo, Milan, Italy  
 Strepparola, Ezio, Bergamo, Italy  
 Scarati, Mario A., Milan, Italy  
 PA Montedison S.p.A., Milan, Italy (non-U.S. corporation)  
 PI US 4721795 19880126  
 AI US 1984-687844 19841231 (6)  
 PRAI IT 1984-21481 19840619  
 DT Utility  
 FS Granted  
 EXNAM Primary Examiner: Chan, Nicky  
 LREP Stevens, Davis, Miller & Mosher  
 CLMN Number of Claims: 3  
 ECL Exemplary Claim: 1  
 DRWN No Drawings  
 LN.CNT 442  
 CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AB Compounds suitable for being used as lubricants, having general formula:

(I)  $\text{RO}-(\text{C}_{\text{sub.3}} \text{F}_{\text{sub.6}} \text{O})_{\text{sub.m}}-(\text{CFXO})_{\text{sub.n}}-\text{CFX}-\text{L}$ , or

(II)  $\text{R}^n(\text{CFXO}-(\text{C}_{\text{sub.3}} \text{F}_{\text{sub.6}} \text{O})_{\text{sub.x}}(\text{CFXO})_{\text{sub.y}}-(\text{C}_{\text{sub.2}} \text{F}_{\text{sub.4}} \text{O})_{\text{sub.z}})-(\text{CFX}-\text{L})$ , where

$\text{R}=\text{CF}_{\text{sub.3}}$ ,  $-\text{C}_{\text{sub.2}} \text{F}_{\text{sub.5}}$ ,  $-\text{C}_{\text{sub.3}} \text{F}_{\text{sub.7}}$

$\text{X}=\text{F}$ ,  $-\text{CF}_{\text{sub.3}}$

$\text{R}^n=\text{F}$ ,  $-\text{CF}_{\text{sub.3}}$ ,  $-\text{C}_{\text{sub.2}} \text{F}_{\text{sub.5}}$

$m$ =an integer from 3 to 100

$n$ =a finite integer, or=zero, wherefore  $m+n$  ranges from 3 to 100, provided that, if  $n$  is finite,  $m/n$  ranges from 5 to 20 and  $\text{R}$  is preferably  $\text{CF}_{\text{sub.3}}$ , if  $n$ =zero,  $\text{R}$  is preferably  $-\text{C}_{\text{sub.2}} \text{F}_{\text{sub.5}}$  or  $-\text{C}_{\text{sub.3}} \text{F}_{\text{sub.7}}$

$x$ =a finite integer, or=zero

$y$ ,  $z$ =finite integers, such that  $x+y+z$  ranges from 5 to 200, while  $x+z/y$  ranges from 5 to 0.5, provided that when  $x$ =zero,  $z/y$  ranges from 1 to 0.5 and  $y+z$  ranges from 5 to 200 while  $\text{X}$  is preferably  $\text{F}$ , and  $\text{R}^n=\text{L}$

$\text{L}$ =group  $\text{A}-\text{Y}$ , where

$\text{A}=-\text{CH}_{\text{sub.2}} \text{O}-$ ,  $-\text{CH}_{\text{sub.2}}-\text{O}-\text{CH}_{\text{sub.2}}$ ,  $-\text{CF}_{\text{sub.2}}$ ,  $\text{CF}_{\text{sub.2}} \text{O}-$ ,

$\text{Y}$ =an organic radical covered by one of the following formulas: ##STR1## where  $\text{R}_{\text{sub.1}}$ ,  $\text{R}_{\text{sub.2}}$ =alkyls  $\text{C}_{\text{sub.1}}-\text{C}_{\text{sub.3}}$ ,

$\text{E}=\text{CHR}_{\text{sub.3}}$  or  $-\text{CH}_{\text{sub.2}}-\text{CHR}_{\text{sub.3}}$

$\text{B}=\text{H}$  or a radical  $\text{OR}_{\text{sub.3}}$  --

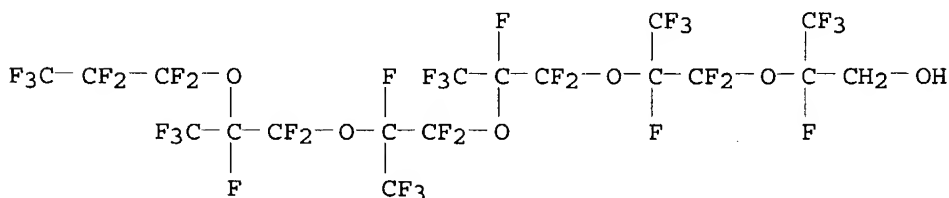
$\text{R}_{\text{sub.3}}=\text{H}$  or an alkyl  $\text{C}_{\text{sub.1}}-\text{C}_{\text{sub.3}}$ .

IT 27617-34-1

(etherification of, by methylenedioxybenzyl chloride)

RN 27617-34-1 USPATFULL

CN 3,6,9,12,15-Pentaoxaoctadecan-1-ol, 2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,17,17,18,18,18-eicosafluoro-2,5,8,11,14-pentakis(trifluoromethyl)- (8CI, 9CI) (CA INDEX NAME)



L21 ANSWER 44 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1986:592200 CAPLUS

DN 105:192200

TI Fluorine-containing polymers with oxygen permeability for medical use

IN Yamauchi, Koichi; Inoue, Yoshihisa; Yokoyama, Kazumasa

PA Green Cross Corp., Japan



10/631,862

SO Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 61111308	A2	19860529	JP 1984-233449	19841106
	JP 05061283	B4	19930906		
PRAI	JP 1984-233449		19841106		

AB The title polymers, useful for hard contact lenses, were prepared from Y(CFXCF2O)mCFX'CH2CH(OH)CH2(OCH2CH2)nO2CCMe:CH2 (X, X' = F, lower perfluoroalkyl; Y = F, lower perfluoroalkoxy; m = 1-8; n = 0, 1) and have number-average mol. weight 700-20,000. Thus, (CF3)2CFO[CF(CF3)CF2O]4CF(CF3)Q

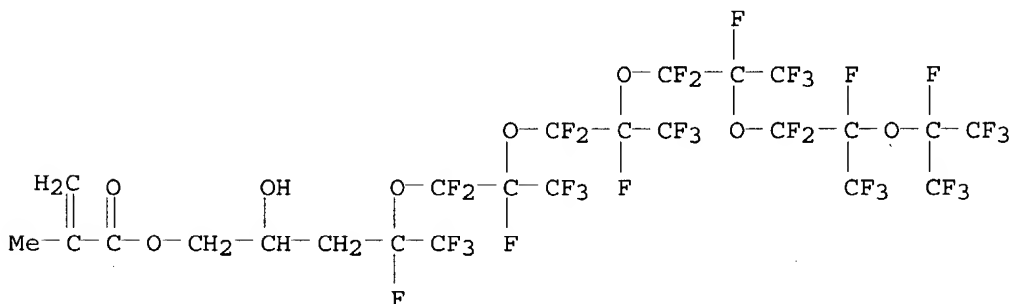
(Q = glycidyl) was copolymd. with trifluoroethyl methacrylate 0.50, vinylpyrrolidone 0.50, Me methacrylate 1.00, benzyl methacrylate 0.80, and allyl methacrylate 0.20 g in the presence of AIBN at 50° for 48 h, at 70° for 5 h, and then at 90° for 3 h to obtain a button which was then dried at 110° for 2 days in vacuo to give Vicat hardness 11 and O permeation 21 + 10-11 cm3-cm/cm2-s-mm Hg.

IT 104937-28-2P

RL: IMF (Industrial manufacture); PREP (Preparation)  
(manufacture and polymerization of)

RN 104937-28-2 CAPLUS

CN 2-Propenoic acid, 2-methyl-, 4,6,6,7,9,9,10,12,12,13,15,15,16,18,19,19,19-heptafluoro-2-hydroxy-4,7,10,13,16,18-hexakis(trifluoromethyl)-5,8,11,14,17-pentaoxanonadec-1-yl ester (9CI) (CA INDEX NAME)



L21 ANSWER 48 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1986:186986 CAPLUS

DN 104:186986

TI Fluoropolyether compounds

IN Caporiccio, Gerardo; Strepparola, Ezio; Scarati, Mario Alberto

PA Montedison S.p.A., Italy

SO Eur. Pat. Appl., 23 pp.

CODEN: EPXXDW

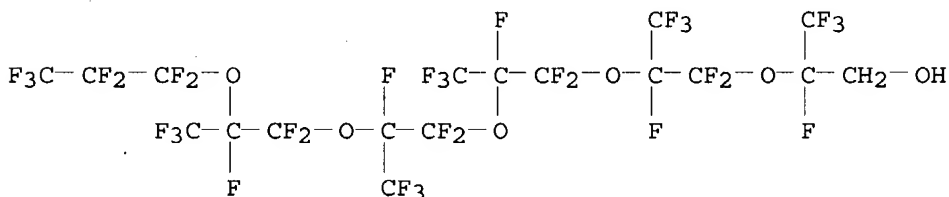
DT Patent

LA English

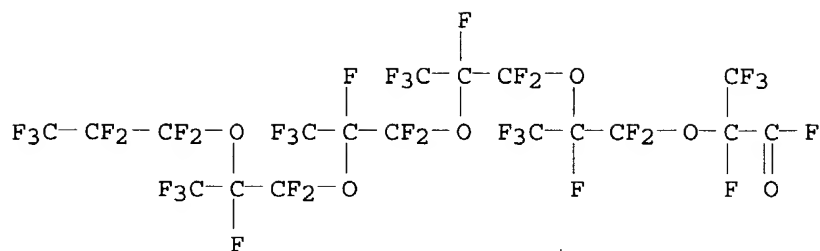
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	EP 165650	A2	19851227	EP 1985-300785	19850206
	EP 165650	A3	19860219		
	EP 165650	B1	19890517		
	R: BE, DE, FR, GB, NL, SE				
	ES 539134	A1	19870501	ES 1984-539134	19841228

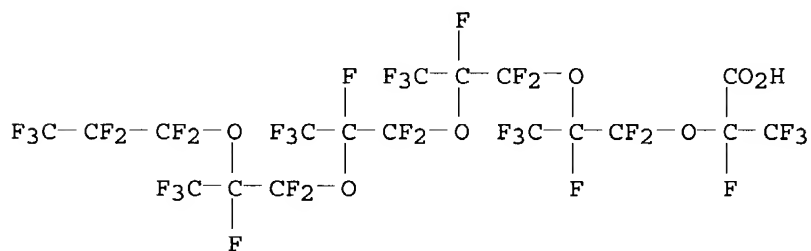
US 4721795 A 19880126 US 1984-687844 19841231  
 CA 1287637 A1 19910813 CA 1985-471406 19850103  
 JP 61004727 A2 19860110 JP 1985-3443 19850114  
 JP 06010257 B4 19940209  
 AU 8537757 A1 19860102 AU 1985-37757 19850117  
 AU 581640 B2 19890302  
 PRAI IT 1984-21481 19840619  
 AB Fluoro polyether compds.  $\text{RO}(\text{C}_3\text{F}_6\text{O})_m(\text{CFXO})_n\text{CFXAY}$  or  $\text{R}_1\text{CFXO}(\text{C}_3\text{F}_6\text{O})_x(\text{CFXO})_y(\text{C}_2\text{F}_4\text{O})_z\text{CFXAY}$  ( $\text{R} = \text{CF}_3, \text{C}_2\text{F}_5, \text{C}_3\text{F}_7$ ;  $\text{X} = \text{F}, \text{CF}_3$ ;  $\text{R}_1 = \text{F}, \text{CF}_3, \text{C}_2\text{F}_5$ ;  $\text{A} = \text{CH}_2\text{O}, \text{CH}_2\text{OCH}_2, \text{CF}_2, \text{CF}_2\text{O}$ ;  $\text{Y} = \text{dialkoxyphenyl}, 1,2\text{-methylenedioxyphenyl}, \text{etc.}$ ) are prepared for use as lubricants or protective coatings for audio or video tapes, floppy disks, etc. Thus, compound I was prepared from  $\text{HOCH}_2\text{CF}_2\text{O}(\text{C}_2\text{F}_4\text{O})_m(\text{CF}_2\text{O})_p\text{CF}_2\text{CH}_2\text{OH}$  ( $m + p = 25$ ,  $m/p = 0.6$ , mol. weight = 2300) 75,  $\text{tert-BuOK}$  8, and 4-chloromethyl-1,2-methylenedioxybenzene 13 g. A 1% solution of I in  $\text{Cl}_2\text{CFCF}_2\text{Cl}$  was coated on a magnetic tape with  $\text{CrO}_2$  pigment. The coating survived 8000 passages of a steel ball (diameter 0.32 mm) with 28 g load in an abrasion test, vs. 350 for a nonlubricated tape.  
 IT 27617-34-1  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (etherification of, by methylenedioxybenzyl chloride)  
 RN 27617-34-1 CAPLUS  
 CN 3,6,9,12,15-Pentaoxaoctadecan-1-ol, 2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,17,17,18,18,18-eicosafuoro-2,5,8,11,14-pentakis(trifluoromethyl)- (8CI, 9CI) (CA INDEX NAME)



L21 ANSWER 49 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 AN 1984:474770 CAPLUS  
 DN 101:74770  
 TI Surface active substances containing an oligo(hexafluoropropene oxide) chain as a hydrophobic oleophobic moiety  
 AU Ishikawa, Nobuo; Sasabe, Mikio  
 CS Dep. Chem. Technol., Tokyo Inst. Technol., Tokyo, 152, Japan  
 SO Journal of Fluorine Chemistry (1984), 25(2), 241-53  
 CODEN: JFLCAR; ISSN: 0022-1139  
 DT Journal  
 LA English  
 AB Oil-soluble surfactants  $\text{CF}_3\text{CF}_2\text{CF}_2\text{O}[\text{CF}(\text{CF}_3)\text{CF}_2\text{O}]_n-2\text{CF}(\text{CF}_3)\text{COR}$  (I) ( $\text{R} = \text{Ph}$  or  $p\text{-tolyl}$ ,  $n = 2-6$ ) were prepared by acylating arenes with hexafluoropropylene oxide oligomers. These surfactants (0.2-0.5%) decreased the surface tensions of toluene and  $m\text{-xylene}$  to 12-14 dynes/cm. Water-soluble surfactants I [ $\text{R} = m\text{-(NaO}_3\text{S)C}_6\text{H}_4$  or  $4\text{-Me-3-(NaO}_3\text{S)C}_6\text{H}_3$ ,  $n = 2-6$ ] were also prepared. Some of the surfactants (i.e.,  $n = 4-6$ ) decreased the surface tension of water to 16 dynes/cm at a concentration of  $10^{-4}$ - $10^{-5}\text{M}$ .  
 IT 13252-15-8  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (acylation by, of arenes)  
 RN 13252-15-8 CAPLUS  
 CN 3,6,9,12,15-Pentaoxaoctadecanoyl fluoride, 2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,17,17,18,18,18-eicosafuoro-2,5,8,11,14-pentakis(trifluoromethyl)- (7CI, 8CI, 9CI) (CA INDEX NAME)



L21 ANSWER 52 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 AN 1983:145452 CAPLUS  
 DN 98:145452  
 TI Synthesis of fluorinated surfactants containing hexafluoropropene oxide as a hydrophobic group and properties of the solutions  
 AU Ogino, Keizo; Murakami, Hiroki; Ishikawa, Nobuo; Sasabe, Mikio  
 CS Fac. Sci. Technol., Sci. Univ. Tokyo, Noda, Japan  
 SO Yukagaku (1983), 32(2), 96-101  
 CODEN: YKGKAM; ISSN: 0513-398X  
 DT Journal  
 LA Japanese  
 AB The surfactants  $\text{C}_3\text{F}_7\text{O}[\text{CF}(\text{CF}_3)\text{CF}_2\text{O}]_n-2\text{CF}(\text{CF}_3)\text{CO}_2\text{Na}$  (I) ( $n = 2-6$ ) were prepared. The critical micelle concentration decreases with increasing  $n$ . A secondary critical micelle concentration is observed for I ( $n = 4-6$ ). The Krafft points of I are  $<0^\circ$ . I ( $n = 4$ ) [67963-78-4] has the best foaming properties. I are stable in acidic and alkaline solns.  
 IT 85248-41-5P  
 RL: SPN (Synthetic preparation); PREP (Preparation) (preparation and surfactant properties of)  
 RN 85248-41-5 CAPLUS  
 CN 3,6,9,12,15-Pentaoxaoctadecanoic acid, 2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,17,17,18,18,18-eicosafuoro-2,5,8,11,14-pentakis(trifluoromethyl)-, sodium salt (9CI) (CA INDEX NAME)



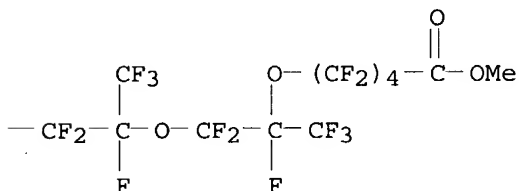
● Na

L21 ANSWER 63 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN  
 AN 1978:508039 CAPLUS  
 DN 89:108039  
 TI Synthesis of perfluoro(polyether) difunctional compounds  
 AU Soloski, E. J.; Tamborski, C.; Psarras, T.

CS Air Force Mater. Lab., Wright-Patterson AFB, OH, USA  
 SO Journal of Fluorine Chemistry (1978), 11(6), 601-12  
 CODEN: JFLCAR; ISSN: 0022-1139  
 DT Journal  
 LA English  
 AB  $\omega$ -Iodoperfluoro(polyether) esters IRfOQfCO<sub>2</sub>R (I; Rf = perfluoroalkylene, Qf = perfluoroalkylene moiety containing 0 atoms in chain, R = Me or Et) were prepared by 2 procedures. I reacted via Zn coupling reactions to give  $\alpha,\omega$ -perfluoro(polyether) diesters. The diesters serve as convenient starting materials for the preparation of a variety of other difunctional compds. of high mol. weight and exhibiting a variation of O-C ratio.  
 IT 61210-96-6P  
 RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)  
 (preparation and amidation of)  
 RN 61210-96-6 CAPLUS  
 CN 6,9,12,15,20,23,26,29-Octaoxatetratriacontanedioic acid,  
 2,2,3,3,4,4,5,5,7,8,8,10,11,11,13,14,14,16,16,17,17,18,18,19,19,21,21,22,2  
 4,24,25,27,27,28,30,30,31,31,32,32,33,33-dotetracontafluoro-  
 7,10,13,22,25,28-hexakis(trifluoromethyl)-, dimethyl ester (9CI) (CA  
 INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

PAGE 1-B



L21 ANSWER 67 OF 108 USPATFULL on STN  
 AN 77:11582 USPATFULL  
 TI Fluoroalkyleneether difunctional compounds  
 IN Tamborski, Christ, Dayton, OH, United States  
 PA The United States of America as represented by the Secretary of the Air  
 Force, Washington, DC, United States (U.S. government)  
 PI US 4011255 19770308  
 AI US 1975-610520 19750904 (5)  
 DT Utility  
 FS Granted  
 EXNAM Primary Examiner: Brust, Joseph Paul  
 LREP Rusz, Joseph E., Kuhn, Cedric H.  
 CLMN Number of Claims: 3  
 ECL Exemplary Claim: 1  
 DRWN No Drawings  
 LN.CNT 304  
 CAS INDEXING IS AVAILABLE FOR THIS PATENT.  
 AB Omega-carbomethoxyperfluoroalkylene ether iodides are reacted with  
 metallic zinc to yield alpha-omega perfluoroalkyleneether diesters. The  
 diesters are reacted with ammonia to form diamides, the diamides are  
 reacted with phosphorus pentoxide to form dinitriles, and the dinitriles  
 are esterified with methanol to form diimide esters. The diimide  
 esters are particularly useful as monomers in synthesizing.

perfluoroalkylene ether bibenzoxazole polymers possessing thermooxidative stability and outstanding low temperature viscoelastic properties.

IT 61210-96-6P

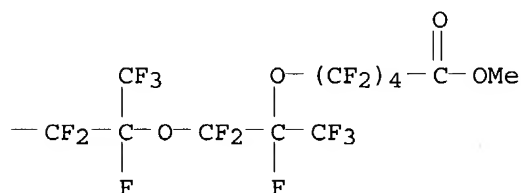
(preparation and amidation of)

RN 61210-96-6 USPATFULL

CN 6,9,12,15,20,23,26,29-Octaoxatetracontanedioic acid,  
2,2,3,3,4,4,5,5,7,8,8,10,11,11,13,14,14,16,16,17,17,18,18,19,19,21,21,22,  
24,24,25,27,27,28,30,30,31,31,32,32,33,33-dotetracontafluoro-  
7,10,13,22,25,28-hexakis(trifluoromethyl)-, dimethyl ester (9CI) (CA  
INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

PAGE 1-B



L21 ANSWER 74 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1976:463699 CAPLUS

DN 85:63699

TI Perfluorinated ethers

IN Von Halasz, Sigmar P.; Kluge, Friedhelm

PA Hoechst A.-G., Fed. Rep. Ger.

SO Ger. Offen., 24 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	DE 2451493	A1	19760506	DE 1974-2451493	19741030
	DE 2451493	C2	19820624		
	NL 7512495	A	19760504	NL 1975-12495	19751024
	US 3985810	A	19761012	US 1975-626349	19751028
	GB 1484823	A	19770908	GB 1975-44343	19751028
	CA 1060482	A1	19790814	CA 1975-238542	19751029
	FR 2289477	A1	19760528	FR 1975-33188	19751030
	FR 2289477	B1	19790105		
PRAI	DE 1974-2451493		19741030		

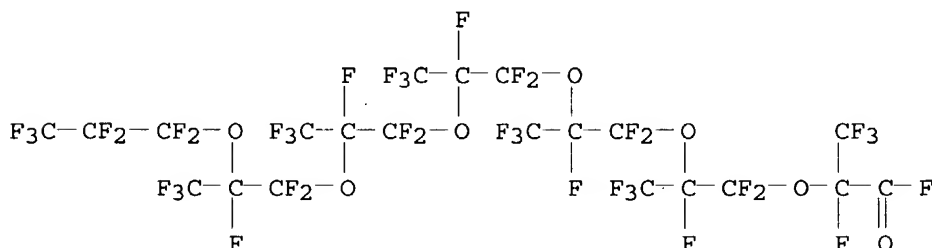
AB The polyethers  $\text{Rf}[[\text{OCF}(\text{R})\text{CF}_2]\text{xOCF}_2\text{R}]_n$  (I) ( $\text{R} = \text{F}, \text{CF}_3$ ;  $\text{Rf} =$  perfluoroalkyl or perfluoroalkylene;  $n = 1-2$ ;  $x = 0-50$ ), useful as hydraulic fluids, heat transfer media, lubricants, etc., are prepared by reaction of F with  $\text{Rf}[[\text{OCF}(\text{R})\text{CF}_2]\text{xOCF}(\text{R})\text{COF}]_n$  (II) in the presence of metal catalysts at  $50-350^\circ$ . Thus, adding 439.5 g II ( $\text{R} = \text{CF}_3$ ,  $\text{Rf} = \text{CF}(\text{CF}_3)\text{CF}(\text{CF}_3)$ ,  $n = 2$ ,  $x = 6.5-9.5$ ) [59859-32-4] over 19.5 hr to a Cu tube packed with silvered Cu filings with countercurrent addition of 0.8 l./hr 3:1 F-He at  $200-5^\circ$  gives 405 g I ( $\text{R} = \text{CF}_3$ ,  $\text{Rf} = \text{CF}(\text{CF}_3)\text{CF}(\text{CF}_3)$ ,  $n = 2$ ,  $x = 13.5-19.5$ ) [59859-33-5], b0.4-0.5  $185-280^\circ$ .

IT 13140-24-4

RL: RCT (Reactant); RACT (Reactant or reagent)  
(fluorination of, to perfluoroalkyl ethers)

10/631,862

RN 13140-24-4 CAPLUS  
CN 3,6,9,12,15,18-Hexaoxaheneicosanoyl fluoride,  
2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,17,19,19,20,20,21,21,21-  
tricosafuoro-2,5,8,11,14,17-hexakis(trifluoromethyl)- (7CI, 8CI, 9CI)  
(CA INDEX NAME)



L21 ANSWER 75 OF 108 USPATFULL on STN  
AN 76:55823 USPATFULL  
TI Process for preparing perfluorinated ethers  
IN von Halasz, Sigmar-Peter, Kelkheim, Taunus, Germany, Federal Republic of  
Kluge, Friedhelm, Frankfurt am Main, Germany, Federal Republic of  
PA Hoechst Aktiengesellschaft, Frankfurt am Main, Germany, Federal Republic  
of (non-U.S. corporation)  
PI US 3985810 19761012  
AI US 1975-626349 19751028 (5)  
PRAI DE 1974-2451493 19741030  
DT Utility  
FS Granted  
EXNAM Primary Examiner: Mars, Howard T.  
LREP Curtis, Morris & Safford  
CLMN Number of Claims: 10  
ECL Exemplary Claim: 1  
DRWN 1 Drawing Figure(s); 1 Drawing Page(s)  
LN.CNT 600  
CAS INDEXING IS AVAILABLE FOR THIS PATENT.  
AB Perfluorinated ethers containing carboxylic acid fluoride groups and  
optionally units derived from hexafluoropropene epoxide or  
tetrafluoroethylene epoxide are reacted with fluorine at temperatures of  
from 50° to 350°C in the presence of metallic catalysts.  
During the reaction carbonyl difluoride is splitt off and an ether is  
obtained in high yield which is free of carboxylic acid fluoride groups.  
Metallic silver is well suited as catalyst.  
IT 13140-24-4  
(fluorination of, to perfluoroalkyl ethers)  
RN 13140-24-4 USPATFULL  
CN 3,6,9,12,15,18-Hexaoxaheneicosanoyl fluoride,  
2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,17,19,19,20,20,21,21,21-  
tricosafuoro-2,5,8,11,14,17-hexakis(trifluoromethyl)- (7CI, 8CI, 9CI)  
(CA INDEX NAME)

[illegible]

AN 1977:4951 CAPLUS

DN 86:4951

IN Tamborski, Christ

SO U. S. Pat. Appl., 14 pp. Avail. NTIS.

CODEN: XAXXAV

DT Patent

LA English

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
1000000	A	1990-01-01	1000000	1990-01-01
1000001	A	1990-01-01	1000001	1990-01-01
1000002	A	1990-01-01	1000002	1990-01-01
1000003	A	1990-01-01	1000003	1990-01-01
1000004	A	1990-01-01	1000004	1990-01-01
1000005	A	1990-01-01	1000005	1990-01-01
1000006	A	1990-01-01	1000006	1990-01-01
1000007	A	1990-01-01	1000007	1990-01-01
1000008	A	1990-01-01	1000008	1990-01-01
1000009	A	1990-01-01	1000009	1990-01-01
1000010	A	1990-01-01	1000010	1990-01-01
1000011	A	1990-01-01	1000011	1990-01-01
1000012	A	1990-01-01	1000012	1990-01-01
1000013	A	1990-01-01	1000013	1990-01-01
1000014	A	1990-01-01	1000014	1990-01-01
1000015	A	1990-01-01	1000015	1990-01-01
1000016	A	1990-01-01	1000016	1990-01-01
1000017	A	1990-01-01	1000017	1990-01-01
1000018	A	1990-01-01	1000018	1990-01-01
1000019	A	1990-01-01	1000019	1990-01-01
1000020	A	1990-01-01	1000020	1990-01-01
1000021	A	1990-01-01	1000021	1990-01-01
1000022	A	1990-01-01	1000022	1990-01-01
1000023	A	1990-01-01	1000023	1990-01-01
1000024	A	1990-01-01	1000024	1990-01-01
1000025	A	1990-01-01	1000025	1990-01-01
1000026	A	1990-01-01	1000026	1990-01-01
1000027	A	1990-01-01	1000027	1990-01-01
1000028	A	1990-01-01	1000028	1990-01-01
1000029	A	1990-01-01	1000029	1990-01-01
1000030	A	1990-01-01	1000030	1990-01-01
1000031	A	1990-01-01	1000031	1990-01-01
1000032	A	1990-01-01	1000032	1990-01-01
1000033	A	1990-01-01	1000033	1990-01-01
1000034	A	1990-01-01	1000034	1990-01-01
1000035	A	1990-01-01	1000035	1990-01-01
1000036	A	1990-01-01	1000036	1990-01-01
1000037	A	1990-01-01	1000037	1990-01-01
1000038	A	1990-01-01	1000038	1990-01-01
1000039	A	1990-01-01	1000039	1990-01-01
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1000043	A	1990-01-01	1000043	1990-01-01
1000044	A	1990-01-01	1000044	1990-01-01
1000045	A	1990-01-01	1000045	1990-01-01
1000046	A	1990-01-01	1000046	1990-01-01
1000047	A	1990-01-01	1000047	1990-01-01
1000048	A	1990-01-01	1000048	1990-01-01
1000049	A	1990-01-01	1000049	1990-01-01
1000050	A	1990-01-01	1000050	1990-01-01
1000051	A	1990-01-01	1000051	1990-01-01
1000052	A	1990-01-01	1000052	1990-01-01
1000053	A	1990-01-01	1000053	1990-01-01
1000054	A	1990-01-01	1000054	1990-01-01
1000055	A	1990-01-01	1000055	1990-01-01
1000056	A	1990-01-01	1000056	1990-0

PI	US 610520	A0	19750904	US 1975-610520	19750904
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PRAI US 1975-610520 19750904

AB MeOC(:NH)(CF<sub>2</sub>)<sub>4</sub>O(CF<sub>2</sub>)<sub>4</sub>O(CF<sub>2</sub>)<sub>4</sub>C(:NH)OMe, a monomer for preparation of elastomeric thermally stable polymers, was prepared by treating ICF<sub>2</sub>CF<sub>2</sub>O(CF<sub>2</sub>)<sub>4</sub>CO<sub>2</sub>Et (I) with Zn to give EtO<sub>2</sub>C(CF<sub>2</sub>)<sub>4</sub>O(CF<sub>2</sub>)<sub>4</sub>O(CF<sub>2</sub>)<sub>4</sub>CO<sub>2</sub>Et, treatment of the diester with NH<sub>3</sub> to give the diamide, dehydration of the diamide with P<sub>2</sub>O<sub>5</sub> to give the dinitrile, and treatment of the latter with Na-MeOH to give the diimidate. MeO<sub>2</sub>C(CF<sub>2</sub>)<sub>4</sub>O[CF(CF<sub>3</sub>)CF<sub>2</sub>O]<sub>n</sub>CF<sub>2</sub>CF<sub>2</sub>I (n = 2, 3) and MeO<sub>2</sub>CCF(CF<sub>3</sub>)OCF<sub>2</sub>CF<sub>2</sub>OCF(CF<sub>3</sub>)CF<sub>2</sub>OCF<sub>2</sub>CF<sub>2</sub>I were also used in place of I.

IT 61210-96-6P

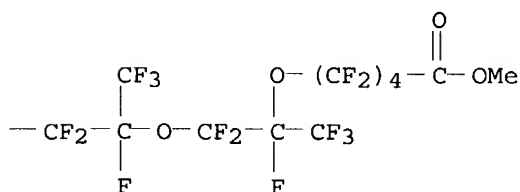
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)  
(preparation and amidation of)

RN 61210-96-6 CAPLUS

CN	6,9,12,15,20,23,26,29-Octaoxatetratriciacontanedioic acid, 2,2,3,3,4,4,5,5,7,8,8,10,11,11,13,14,14,16,16,17,17,18,18,19,19,21,21,22,2 4,24,25,27,27,28,30,30,31,31,32,32,33,33-dotetracontafluoro- 7,10,13,22,25,28-hexakis(trifluoromethyl)-, dimethyl ester (9CI) . (CA . INDEX NAME)
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\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

PAGE 1-B



L21 ANSWER 88 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1975:594115 CAPLUS

DN 83:194115

TI Perfluorinated linear polyethers having reactive terminal groups at both ends of the chain

IN Sianesi, Dario; Caporiccio, Gerardo; Mensi, Domenico

PA Montedison S.p.A., Italy

SO U.S., 14 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 3847978	A	19741112	US 1969-834486	19690618
PRAI	US 1968-787309		19681226		

AB Perfluorinated linear polyethers containing peroxidic linkages were chain-cleaved by reducing agents to give bifunctional perfluorinated linear oligopolyethers with chemical-reactive terminal groups. Thus, hexafluoropropene [116-15-4] was treated with oxygen under the influence of uv light to give a peroxidized poly(perfluoropropylene oxide) [25038-02-2] which was reduced by H over a Pd catalyst to give a series of carboxy- and trifluoroacetyl-terminated oligopolyethers. One of these, CF<sub>3</sub>COCF<sub>2</sub>O(CF<sub>3</sub>CF<sub>2</sub>)<sub>2</sub>CF(CF<sub>3</sub>)CO<sub>2</sub>H [42775-40-6], boiling point 210-2°, formed a polymer with hexamethylenediamine [55809-69-3].

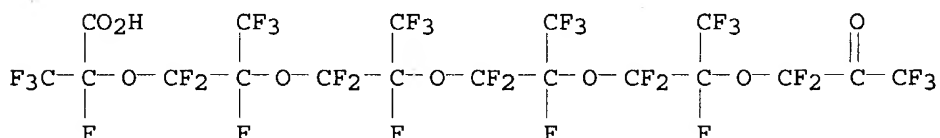
IT 42775-42-8P

RL: IMF (Industrial manufacture); PREP (Preparation)

(manufacture of, by reduction of perfluorinated polyether peroxy derivs.)

RN 42775-42-8 CAPLUS

CN 3,6,9,12,15-Pentaoxaooctadecanoic acid, 2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,18,18,18-octadecafluoro-17-oxo-2,5,8,11,14-pentakis(trifluoromethyl)-(9CI) (CA INDEX NAME)



L21 ANSWER 89 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1975:88056 CAPLUS

DN 82:88056

TI Perfluoroalkyletheramidoalkyl betaines and sulfobetaines

IN Barlett, Phillip Lee

PA du Pont de Nemours, E. I., and Co.

SO U.S., 4 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 3839425	A	19741001	US 1970-72803	19700916
PRAI	US 1970-72803		19700916		

AB Thirteen surfactants [C<sub>3</sub>F<sub>7</sub>O[CF(CF<sub>3</sub>)CF<sub>2</sub>O]<sub>n</sub>CF(CF<sub>3</sub>)CONR(CH<sub>2</sub>)<sub>m</sub>N<sup>+</sup>(R<sub>1</sub>)<sub>2</sub>R<sub>2</sub> with R = H, Me, or Et, R<sub>1</sub> = Me, Et, or Pr, R<sub>2</sub> = (CH<sub>2</sub>)<sub>1-2</sub>CO<sub>2</sub><sup>-</sup> or (CH)<sub>2</sub>-SO<sub>3</sub><sup>-</sup>, m = 2-3, and n = 0-4] were prepared and were especially useful as foam stabilizers for



10/631,862

fire-extinguishing foams on burning hydrocarbon surfaces. Thus, 50 g C3F7OCF(CF3)CF2OCF(CF3)CONH(CH2)3NMe2 [31339-59-0], 10.1 g ClCH2CO2Na [3926-62-3], and 20 ml iso-PrOH were refluxed for 16 hr to prepare 53.7 g C3F7OCF(CF3)CF2OCF(CF3)CONH(CH2)3N+Me2CH2CO2- [54190-98-6] which gave surface tension 18.7 dynes/cm as a 0.001% aqueous solution and, as a 1% solution,

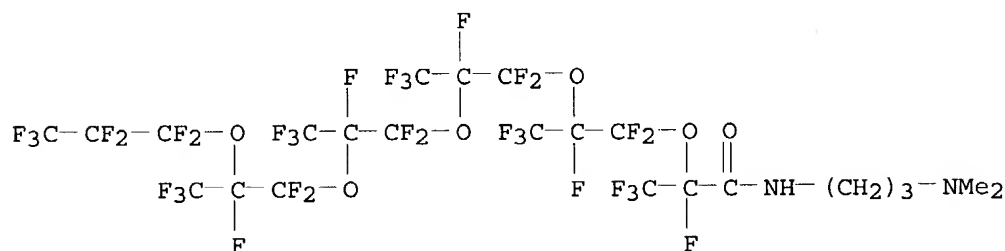
caused water to spread rapidly over the surface of cyclohexane.

IT 54190-86-2

RL: RCT (Reactant); RACT (Reactant or reagent)  
(reaction of, with carboxyalkyl and sulfoalkyl halides)

RN 54190-86-2 CAPLUS

CN 3,6,9,12,15-Pentaoxaoctadecanamide, N-[3-(dimethylamino)propyl]-  
2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,17,17,18,18,18-eicosafuoro-  
2,5,8,11,14-pentakis(trifluoromethyl)- (9CI) (CA INDEX NAME)



L21 ANSWER 100 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1972:16000 CAPLUS

DN 76:16000

TI Perfluoroalkyl ether amidoamine oxides

IN Bartlett, Philip L.

PA du Pont de Nemours, E. I., and Co.

SO U.S., 6 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 3

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI	US 3547995	A	19701215	US 1968-705932	19680216
	NL 6803275	A	19680909	NL 1968-3275	19680307
	FR 1568163	A	19690523	FR 1968-1568163	19680307
	GB 1202830	A	19700819	GB 1968-1202830	19680307
	DE 1793761	A1	19730823	DE 1967-1793761	19680307
PRAI	US 1967-621128		19670307		
	US 1967-621148		19670307		
	US 1967-621157		19670307		
	US 1968-705923		19680216		
	US 1968-705932		19680216		
	US 1968-705947		19680216		

AB A group of perfluoroalkyl ether amidoamine oxides are useful as surface active agents and are noncorrosive to steel. Hexafluoropropylene oxide is trimerized to perfluoroalkyl ether acid fluoride which is esterified with methanol and reacted with 3-(dimethylamino)propylamine to give a corresponding perfluoro alkyl ether amide which was oxidized to [3-[2-[2-(heptafluoropropoxy)hexafluoropropoxy]tetrafluoropropionamidol]propyl]dimethylamine oxide [29209-86-7].

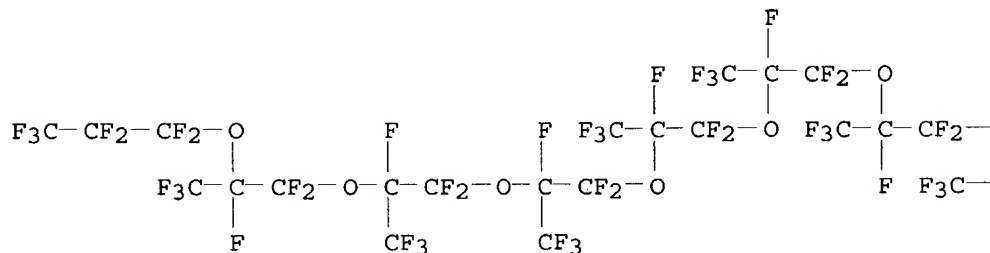
IT 34839-72-0

RL: TEM (Technical or engineered material use); USES (Uses)  
(surfactants)

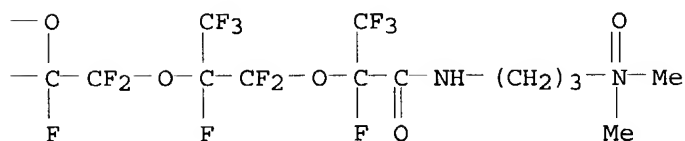
RN 34839-72-0 CAPLUS

CN 3,6,9,12,15,18,21,24-Octaoxatriacontanamide, N-[3-(dimethyloxidoamino)propyl]-2,4,4,5,7,7,8,10,10,11,13,13,14,16,16,17,19,19,20,22,22,23,25,25,26,28,28,29,29,30,30,30-dotriacontafluoro-2,5,8,11,14,17,20,23,26-nonakis(trifluoromethyl)- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 1-B



L21 ANSWER 106 OF 108 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1968:77639 CAPLUS

DN 68:77639

TI Perfluorinated polyethers. Synthesis and characterization of a new class of inert fluids

AU Sianesi, Dario

CS "Montecatini Edison", Ist. "G. Donegani", Milan-Linate, Italy

SO Chimica e l'Industria (Milan, Italy) (1968), 50(2), 206-14

CODEN: CINMAB; ISSN: 0009-4315

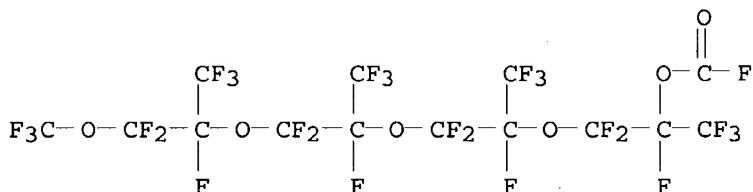
DT Journal

LA Italian

AB The photochem. reaction between hexafluoropropylene and O was studied. Compds. of the general formulas  $\text{CF}_3(\text{OR})_n\text{O}[\text{CF}_2\text{CF}(\text{CF}_3)]_m\text{COF}$  (I),  $\text{CF}_3(\text{OR})_n\text{O}[\text{CF}_2\text{CF}(\text{CF}_3)\text{O}]_m\text{CF}_2\text{COCF}_3$  (II), and  $\text{CF}_3(\text{OR})_n\text{O}[\text{CF}_2\text{CF}(\text{CF}_3)\text{O}]_m\text{CF}_2\text{H}$  (III) are obtained. The following I (n, R, m, and b.p. given): 0, -, 0, -; 0, -, 1, 51°; 0, -, 2, 114°; 0, -, 3, 156-7°; 0, -, 4, 195-7°; 1,  $\text{CF}_2$ , 1, 83-6°; 1,  $\text{CF}_2$ , 2, 130-3°; 1,  $\text{CF}(\text{CF}_2)$ , 1, 96-8°; 1,  $\text{CF}(\text{CF}_3)$ , 2, 143-5°; the following II (n, R, m, and b.p. given): 0, -, 0, 15°; 0, -, 1, 87°; 0, -, 2, 137°; 0, -, 3, 180-1°; 0, -, 4, 215-16°; 0, -, 5, 244-5°; 1,  $\text{CF}_2$ , 0, 48-9°; 1,  $\text{CF}_2$ , 1, 106-7°; 1,  $\text{CF}_2$ , 2, 157-60°; 1,  $\text{CF}_2$ , 3, 197-200°; 1,  $\text{CF}(\text{CF}_3)$ , 0, 68°; 1,  $\text{CF}(\text{CF}_3)$ , 1, 121-2°; 1,  $\text{CF}(\text{CF}_3)$ , 2, 168-70°; 1,  $\text{CF}(\text{CF}_3)$ , 3, 205-7°; and the following III (n, R, m, and b.p. given): 1, -, 0, -36°; 0, -, 1, 55°; 0, -, 2, 113°; 0, -, 3, 161-2°; 0, -, 4, 200-1°; 0, -, 5, 231-2°; 1,  $\text{CF}_2$ , 1, 88-90°; 1,  $\text{CF}_2$ , 2, 133-4°; 1,  $\text{CF}_2$ , 3, 175-8°; 1,  $\text{CF}(\text{CF}_3)$ , 1, 100-1°; 1,  $\text{CF}(\text{CF}_3)$ , 2, 147-8°; 1,  $\text{CF}(\text{CF}_3)$ , 3, 189-92°, are prepared

10/631,862

IT 18934-94-6P  
RL: SPN (Synthetic preparation); PREP (Preparation)  
(preparation of)  
RN 18934-94-6 CAPLUS  
CN Formic acid, fluoro-, trifluoro-2-[trifluoro-1-(trifluoromethyl)-2-  
[trifluoro-1-(trifluoromethyl)-2-[trifluoro-2-(trifluoromethoxy)-1-  
(trifluoromethyl)ethoxy]ethoxy]ethoxy]-1-(trifluoromethyl)ethyl ester  
(8CI) (CA INDEX NAME)



=> file stnguide

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	152.55	469.93

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE ENTRY	TOTAL SESSION
CA SUBSCRIBER PRICE	-9.10	-9.10

FILE 'STNGUIDE' ENTERED AT 07:28:50 ON 10 NOV 2004  
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AND TECHNOLOGY CORPORATION, AND FACHINFORMATIONSZENTRUM KARLSRUHE

FILE CONTAINS CURRENT INFORMATION.  
LAST RELOADED: Nov 5, 2004 (20041105/UP).

=> file reg

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	0.18	470.11

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE ENTRY	TOTAL SESSION
CA SUBSCRIBER PRICE	0.00	-9.10

FILE 'REGISTRY' ENTERED AT 07:30:31 ON 10 NOV 2004  
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Property values tagged with IC are from the ZIC/VINITI data file  
provided by InfoChem.

STRUCTURE FILE UPDATES: 8 NOV 2004 HIGHEST RN 777024-10-9  
DICTIONARY FILE UPDATES: 8 NOV 2004 HIGHEST RN 777024-10-9

TSCA INFORMATION NOW CURRENT THROUGH MAY 21, 2004

Please note that search-term pricing does apply when  
conducting SmartSELECT searches.

10/631,862

Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. For more information enter HELP PROP at an arrow prompt in the file or refer to the file summary sheet on the web at:  
<http://www.cas.org/ONLINE/DBSS/registryss.html>

=> ....Testing the current file.... screen

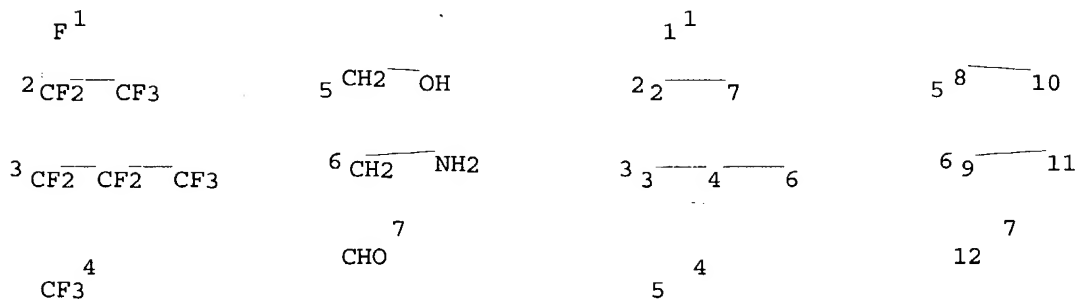
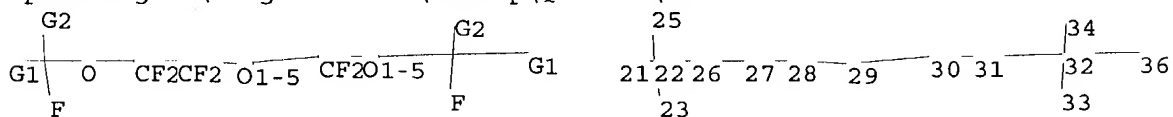
ENTER SCREEN EXPRESSION OR (END):end

=> screen 1994 OR 2016 OR 2021 OR 2026 OR 1838

L22 SCREEN CREATED

=>

Uploading C:\Program Files\Stnexp\Queries\10630698b-1.str



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 21 22 23 25 26 27 28 29 30 31 32  
33 34 36

chain bonds :

2-7 3-4 4-6 8-10 9-11 21-22 22-23 22-25 22-26 26-27 27-28 28-29 29-30  
30-31 31-32 32-36 32-33 32-34

exact/norm bonds :

10/631,862

21-22 22-25 22-26 31-32 32-36 32-34

exact bonds :

2-7 3-4 4-6 8-10 9-11 22-23 26-27 27-28 28-29 29-30 30-31 32-33

G1:[\*1],[\*2],[\*3],[\*4],[\*5],[\*6],[\*7]

G2:[\*1],[\*2]

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS  
10:CLASS 11:CLASS 12:CLASS 21:CLASS 22:CLASS 23:CLASS 25:CLASS 26:CLASS  
27:CLASS 28:CLASS 29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS  
36:CLASS

L23 STRUCTURE UPLOADED

=> que L23 NOT L22

L24 QUE L23 NOT L22

=> s l24

SAMPLE SEARCH INITIATED 07:31:27 FILE 'REGISTRY'

SAMPLE SCREEN SEARCH COMPLETED - 265 TO ITERATE

100.0% PROCESSED 265 ITERATIONS ( 4 INCOMPLETE) 6 ANSWERS  
SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE \*\*COMPLETE\*\*  
BATCH \*\*COMPLETE\*\*

PROJECTED ITERATIONS: 4324 TO 6276  
PROJECTED ANSWERS: 6 TO 266

L25 6 SEA SSS SAM L23 NOT L22

=> s l24 ful

FULL SEARCH INITIATED 07:31:35 FILE 'REGISTRY'

FULL SCREEN SEARCH COMPLETED - 4703 TO ITERATE

100.0% PROCESSED 4703 ITERATIONS ( 70 INCOMPLETE) 81 ANSWERS  
SEARCH TIME: 00.00.03

L26 81 SEA SSS FUL L23 NOT L22

=> file caplus casreact uspatful

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	156.26	626.37

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE ENTRY	TOTAL SESSION
CA SUBSCRIBER PRICE	0.00	-9.10

FILE 'CAPLUS' ENTERED AT 07:32:23 ON 10 NOV 2004

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FILE 'CASREACT' ENTERED AT 07:32:23 ON 10 NOV 2004

10/631,862 .

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FILE 'USPATFULL' ENTERED AT 07:32:23 ON 10 NOV 2004  
CA INDEXING COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

=> s 126

L27 76 L26

=> dup rem 127

PROCESSING COMPLETED FOR L27

L28 69 DUP REM L27 (7 DUPLICATES REMOVED)

=> d 1-69 ti

L28 ANSWER 1 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Manufacture of magnetic recording media

L28 ANSWER 2 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Perfluoropoly-ether/peroxide compounds: spectroscopic studies and quantum chemical calculations

L28 ANSWER 3 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Fluorine-containing compounds, lubricants and magnetic recording media therewith, and manufacture thereof

L28 ANSWER 4 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Manufacture of magnetic recording media

L28 ANSWER 5 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Fluorine-containing tertiary amine tricarboxylate ester, lubricant, magnetic recording medium using the lubricant, and manufacture of the recording medium

L28 ANSWER 6 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Magnetic recording media having good traveling durability and high electromagnetic conversion and their manufacture

L28 ANSWER 7 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Perfluoroalkyl polyether oligomers containing phosphazene groups useful as lubricants for recording media such as hard disks

L28 ANSWER 8 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Magnetic recording medium and its fabrication

L28 ANSWER 9 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Thin magnetic tapes with good durability having stainless reinforcing layers on their back side and their manufacture

L28 ANSWER 10 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Magnetic tapes with fluorine-containing lubricant layers and their manufacture

L28 ANSWER 11 OF 69 USPATFULL on STN

TI Rolling bearing

L28 ANSWER 12 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI (Fluoroorgano)silicon compounds as hydro- and oleophobic agents for protection of building materials from adverse effects of environment

L28 ANSWER 13 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Lubricating grease for sliding bearings

- L28 ANSWER 14 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Monoesters of fluorinated alkyldicarboxylic acids, lubricant compositions, magnetic recording media, and their manufacture
- L28 ANSWER 15 OF 69 USPATFULL on STN  
TI Magnetic recording medium having a perfluoropolyether lubricant bonded to the surface of a carbon protective film
- L28 ANSWER 16 OF 69 USPATFULL on STN  
TI Liquid-phase fluorination
- L28 ANSWER 17 OF 69 USPATFULL on STN  
TI Liquid-phase fluorination
- L28 ANSWER 18 OF 69 USPATFULL on STN  
TI Amides and esters of perfluoropolyoxaalkylene-sulfo- or perfluoropolyoxaalkylene-carboxylic acids and a process for producing same
- L28 ANSWER 19 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 1  
TI Method of manufacturing a magnetic storage medium
- L28 ANSWER 20 OF 69 USPATFULL on STN  
TI Liquid phase fluorination
- L28 ANSWER 21 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Synthesis of 1,1-dihydroperfluorooxaalkan-1-ols and their reaction with terephthaloyl chloride
- L28 ANSWER 22 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI The solid-like state of a confined liquid lubricant: deformation and time effects
- L28 ANSWER 23 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI The effect of adhesion on the rheological and frictional behavior of a confined lubricant film
- L28 ANSWER 24 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Manufacture of amides and esters of perfluoropolyoxyalkylenesulfonic or -carboxylic acids
- L28 ANSWER 25 OF 69 USPATFULL on STN  
TI Liquid phase fluorination
- L28 ANSWER 26 OF 69 USPATFULL on STN  
TI Curing fluorocarbon elastomers
- L28 ANSWER 27 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Cocyclotrimerization of mono- and dinitriles of perfluorocarboxylic acids under high pressure
- L28 ANSWER 28 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Development of quantitative structure-activity relationships for perfluoropolyalkyl ethers
- L28 ANSWER 29 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 2  
TI Liquid-phase fluorination
- L28 ANSWER 30 OF 69 USPATFULL on STN  
TI Liquid-phase fluorination

10/631,862

- L28 ANSWER 31 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 3  
TI Features of cyclotrimerization of perfluoroalkyl- and perfluorooxaalkylacetylenes
- L28 ANSWER 32 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI From static to kinetic friction in confined liquid films
- L28 ANSWER 33 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI New fluorinated oligomers and polymers based on (perfluoroalkyl)- and (perfluorooxyalkylene)acetylenes
- L28 ANSWER 34 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Lubricants for magnetic recording medium
- L28 ANSWER 35 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Curable fluororubber compositions containing fluorinated polyethers
- L28 ANSWER 36 OF 69 USPATFULL on STN  
TI Perfluorinated polyethers and process for their preparation
- L28 ANSWER 37 OF 69 USPATFULL on STN  
TI Liquid phase fluorination
- L28 ANSWER 38 OF 69 USPATFULL on STN  
TI Novel perfluorinated polyethers and process for their preparation
- L28 ANSWER 39 OF 69 USPATFULL on STN  
TI Novel perfluorinated polyethers and process for their preparation
- L28 ANSWER 40 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Colloid chemical properties of aqueous solutions of derivatives of perfluorooxaalkylcarboxylic acids based on oligomers of tetrafluoroethylene oxide
- L28 ANSWER 41 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Perfluoro tertiary alcohols. I. Synthesis of high molecular weight perfluorinated monoketones and tertiary alcohols
- L28 ANSWER 42 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Preparation of perfluoroacetal and perfluoroketal compounds and use thereof in thermal shock testing
- L28 ANSWER 43 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Perfluorination of hydrogen-containing compounds
- L28 ANSWER 44 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Perfluorination of alcohol ethoxylates
- L28 ANSWER 45 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 4  
TI Preparation of carbonyl fluoride compounds
- L28 ANSWER 46 OF 69 USPATFULL on STN  
TI Perfluoro-keto-ylids as precursors of polychloroketones, 1,2-diketones and quinoxalines
- L28 ANSWER 47 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 5  
TI Fluoroacrylate polymers and copolymers for manufacture of contact lenses.
- L28 ANSWER 48 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
TI Lubricating oils for refrigerating compressors
- L28 ANSWER 49 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN



TI Manufacture of magnetic memory disks

L28 ANSWER 50 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Optical pickup activator

L28 ANSWER 51 OF 69 USPATFULL on STN

TI Perfluoro-keto-ylids as precursors of polychloroketones, 1,2-diketones and quinoxalines

L28 ANSWER 52 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 6

TI Perfluoro-keto-ylids as precursors of polychloroketones, 1,2-diketones and quinoxalines

L28 ANSWER 53 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 7

TI Perfluorocarbon ethers from a high-molecular-weight polyether

L28 ANSWER 54 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Magnetic recording medium having a perfluoropolyether polymer protective coating

L28 ANSWER 55 OF 69 USPATFULL on STN

TI Magnetic recording medium having a perfluoropolyether polymer protective coating

L28 ANSWER 56 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Fluoroalkylene ether silicate/viton GLT blends: an approach toward improved low temperature flexibility

L28 ANSWER 57 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Hydrolytically stable fluorocarbon ether bibenzoxazole polymers

L28 ANSWER 58 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI F-Phenylalkylene oxide diacetylenes

L28 ANSWER 59 OF 69 USPATFULL on STN

TI F-Phenylalkylene oxide diacetylenes

L28 ANSWER 60 OF 69 USPATFULL on STN

TI Fluoroalkyleneether silicate copolymers

L28 ANSWER 61 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Fluoroalkyleneether silicate copolymers

L28 ANSWER 62 OF 69 USPATFULL on STN

TI Hybrid perfluoroalkylene ether thioimide ester monomers

L28 ANSWER 63 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Synthesis of the perfluoropoly(ethylene glycol) ethers by direct fluorination

L28 ANSWER 64 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Hybrid perfluoroalkylene ether thioimide ester monomers

L28 ANSWER 65 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Synthesis of perfluoropoly(ethylene glycol) ethers  $\text{CF}_3[\text{OCF}_2\text{CF}_2]_n\text{ORf}$  ( $\text{Rf} = \text{CF}_3$  or  $\text{C}_2\text{F}_5$ ;  $n = 1-5$ )

L28 ANSWER 66 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI The addition of tetrafluoroethylene oxide to F-glutaryl fluoride. Relative reactivities of acid fluorides

L28 ANSWER 67 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

10/631,862

TI Nature of the addition of perfluoroolefin oxides to perfluorodicarboxylic acid difluorides

L28 ANSWER 68 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Oil repellent polyfluoropolyoxo-alkyl phosphates

L28 ANSWER 69 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

TI Fluorocarbon ethers of tetrafluoroethylene epoxide

=> 2,16,17,20,25,29,30,36,37,38,39,44,45,53,65,66,67 bib ab fhitr  
2,16,17,20,25,29,30,36,37,38,39,44,45,53,65,66,67 IS NOT A RECOGNIZED COMMAND  
The previous command name entered was not recognized by the system.  
For a list of commands available to you in the current file, enter  
"HELP COMMANDS" at an arrow prompt (=>).

=> d 2,16,17,20,25,29,30,36,37,38,39,44,45,53,65,66,67 bib ab fhitr

L28 ANSWER 2 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

AN 2004:75006 CAPLUS

DN 140:375766

TI Perfluoropoly-ether/peroxide compounds: spectroscopic studies and quantum chemical calculations

AU Radice, S.; Toniolo, P.; Barchiesi, E.; Guarda, P. A.; Tommasini, M.; Castiglioni, C.

CS Solvay Solexis, Bollate (MI), 20021, Italy

SO Journal of Fluorine Chemistry (2004), 125(2), 151-164

CODEN: JFLCAR; ISSN: 0022-1139

PB Elsevier Science B.V.

DT Journal

LA English

AB Perfluoropolyethers (PFPEs) are a class of high performance materials used in a wide range of applications (refrigeration, lubrication, semiconductor industry, etc.). PFPEs containing peroxidic units are intermediate materials for the preparation of com. end products. In this work we study the spectroscopic properties of ether and peroxides linkages in this class of compds.; NMR (NMR) spectra are discussed, FT-Raman data presented. Quantum chemical calcns. on model mols. were used as a tool for the interpretation of the Raman exptl. data and phys.-chemical properties.

IT 67584-24-1

RL: PRP (Properties)

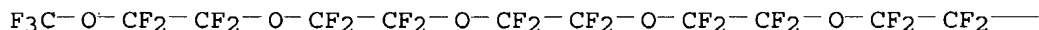
(model compound; spectroscopic studies and quantum chemical calcns.

perfluoropolyether/peroxide compds. prepared by oxidative polymerization of tetrafluoroethylene)

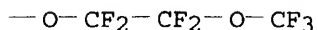
RN 67584-24-1 CAPLUS

CN 2,5,8,11,14,17,20-Heptaohexacosane, 1,1,1,3,3,4,4,6,6,7,7,9,9,10,10,12,12,13,13,15,15,16,16,18,18,19,19,21,21,21-triacontafuoro- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 1-B

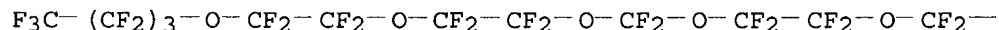


RE.CNT 26 THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS RECORD

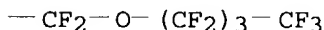
## ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 16 OF 69 USPATFULL on STN  
 AN 1998:55015 USPATFULL  
 TI Liquid-phase fluorination  
 IN Bierschenk, Thomas R., Round Rock, TX, United States  
 Juhlke, Timothy J., Round Rock, TX, United States  
 Kawa, Hajimu, Austin, TX, United States  
 Lagow, Richard J., Austin, TX, United States  
 PA Exfluor Research Corporation, Round Rock, TX, United States (U.S. corporation)  
 PI US 5753776 19980519  
 AI US 1995-471031 19950606 (8)  
 RLI Continuation of Ser. No. US 1994-258708, filed on 13 Jun 1994, now patented, Pat. No. US 5461117, issued on 24 Oct 1995 which is a continuation of Ser. No. US 1993-28721, filed on 8 Mar 1993, now patented, Pat. No. US 5322904, issued on 21 Jun 1994 which is a continuation-in-part of Ser. No. US 1992-823837, filed on 17 Jan 1992, now abandoned which is a continuation of Ser. No. US 1989-414119, filed on 28 Sep 1989, now patented, Pat. No. US 5093432, issued on 3 Mar 1992 which is a continuation-in-part of Ser. No. US 1988-250376, filed on 28 Sep 1988, now abandoned  
 DT Utility  
 FS Granted  
 EXNAM Primary Examiner: Krass, Frederick  
 LREP Hamilton, Brook, Smith & Reynolds, P.C.  
 CLMN Number of Claims: 35  
 ECL Exemplary Claim: 1  
 DRWN 2 Drawing Figure(s); 2 Drawing Page(s)  
 LN.CNT 2151  
 CAS INDEXING IS AVAILABLE FOR THIS PATENT.  
 AB This invention pertains to a method for liquid phase fluorination for perfluorination of a wide variety of hydrogen-containing compounds.  
 IT 130085-03-9P  
 (preparation of)  
 RN 130085-03-9 USPATFULL  
 CN 5,8,11,13,16,19-Hexaoxatricosane, 1,1,1,2,2,3,3,4,4,6,6,7,7,9,9,10,10,12,12,14,14,15,15,17,17,18,18,20,20,21,21,22,22,23,23,23-hexatriacontafluoro-(9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 1-B



L28 ANSWER 17 OF 69 USPATFULL on STN  
 AN 97:91604 USPATFULL  
 TI Liquid-phase fluorination  
 IN Bierschenk, Thomas R., Round Rock, TX, United States  
 Juhlke, Timothy, Round Rock, TX, United States  
 Kawa, Hajimu, Austin, TX, United States  
 Lagow, Richard J., Austin, TX, United States  
 PA Exfluor Research Corporation, Round Rock, TX, United States (U.S. corporation)

10/631,862

PI US 5674949 19971007  
AI US 1995-466798 19950606 (8)  
RLI Continuation of Ser. No. US 1994-240225, filed on 10 May 1994, now patented, Pat. No. US 5571870, issued on 5 Nov 1996 which is a continuation of Ser. No. US 1992-823836, filed on 17 Jan 1992, now patented, Pat. No. US 5322903, issued on 21 Jun 1994 which is a continuation of Ser. No. US 1989-414119, filed on 28 Sep 1989, now patented, Pat. No. US 5093432, issued on 3 Mar 1992 which is a continuation-in-part of Ser. No. US 1988-250376, filed on 28 Sep 1988, now abandoned  
DT Utility  
FS Granted  
EXNAM Primary Examiner: Krass, Frederick  
LREP Hamilton, Brook, Smith & Reynolds, P.C.  
CLMN Number of Claims: 20  
ECL Exemplary Claim: 1  
DRWN 2 Drawing Figure(s); 2 Drawing Page(s)  
LN.CNT 2088  
CAS INDEXING IS AVAILABLE FOR THIS PATENT.  
AB This invention pertains to a method for liquid-phase fluorination for perfluorination of a wide variety of hydrogen-containing compounds.  
IT 130085-03-9P  
(preparation of)  
RN 130085-03-9 USPATFULL  
CN 5,8,11,13,16,19-Hexaoxatricosane, 1,1,1,2,2,3,3,4,4,6,6,7,7,9,9,10,10,12,12,14,14,15,15,17,17,18,18,20,20,21,21,22,22,23,23,23-hexatriacontafluoro-  
(9CI) (CA INDEX NAME)

PAGE 1-A

$\text{F}_3\text{C}-(\text{CF}_2)_3-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-$

PAGE 1-B

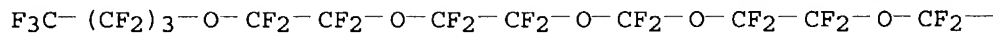
$-\text{CF}_2-\text{O}-(\text{CF}_2)_3-\text{CF}_3$

L28 ANSWER 20 OF 69 USPATFULL on STN  
AN 96:101634 USPATFULL  
TI Liquid phase fluorination  
IN Bierschenk, Thomas R., Round Rock, TX, United States  
Juhlke, Timothy, Round Rock, TX, United States  
Kawa, Hajimu, Austin, TX, United States  
Lagow, Richard J., Austin, TX, United States  
PA Exfluor Research Corporation, Round Rock, TX, United States (U.S. corporation)  
PI US 5571870 19961105  
AI US 1994-240225 19940510 (8)  
DCD 20110621  
RLI Continuation of Ser. No. US 1992-823836, filed on 17 Jan 1992, now patented, Pat. No. US 5322903, issued on 21 Jun 1994 which is a continuation of Ser. No. US 1989-414119, filed on 28 Sep 1989, now patented, Pat. No. US 5093432, issued on 3 Mar 1992 which is a continuation-in-part of Ser. No. US 1988-250376, filed on 28 Sep 1988, now abandoned  
DT Utility  
FS Granted  
EXNAM Primary Examiner: Krass, Frederick

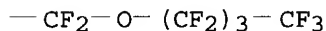
10/631,862

LREP Hamilton, Brook, Smith & Reynolds, P.C.  
CLMN Number of Claims: 28  
ECL Exemplary Claim: 1  
DRWN 2 Drawing Figure(s); 2 Drawing Page(s)  
LN.CNT 2065  
CAS INDEXING IS AVAILABLE FOR THIS PATENT.  
AB This invention pertains to a method for liquid phase fluorination for perfluorination of a wide variety of hydrogen-containing compounds.  
IT 130085-03-9P  
(preparation of)  
RN 130085-03-9 USPATFULL  
CN 5,8,11,13,16,19-Hexaoxatricosane, 1,1,1,2,2,3,3,4,4,6,6,7,7,9,9,10,10,12,12,14,14,15,15,17,17,18,18,20,20,21,21,22,22,23,23,23-hexatriacontafluoro-  
(9CI) (CA INDEX NAME)

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L28 ANSWER 25 OF 69 USPATFULL on STN  
AN 95:94983 USPATFULL  
TI Liquid phase fluorination  
IN Bierschenk, Thomas R., Round Rock, TX, United States  
Juhlke, Timothy J., Round Rock, TX, United States  
Kawa, Hajimu, Austin, TX, United States  
Lagow, Richard J., Austin, TX, United States  
PA Exfluor Research Corporation, Austin, TX, United States (U.S. corporation)  
PI US 5461117 19951024  
AI US 1994-258708 19940613 (8)  
RLI Continuation of Ser. No. US 1993-28721, filed on 8 Mar 1993, now patented, Pat. No. US 5322904 which is a continuation-in-part of Ser. No. US 1992-823837, filed on 17 Jan 1992, now abandoned which is a continuation of Ser. No. US 1989-414119, filed on 28 Sep 1989, now patented, Pat. No. US 5094432, issued on 3 Mar 1992 which is a continuation-in-part of Ser. No. US 1988-250376, filed on 28 Sep 1988, now abandoned  
DT Utility  
FS Granted  
EXNAM Primary Examiner: Krass, Frederick  
LREP Hamilton, Brook, Smith & Reynolds  
CLMN Number of Claims: 30  
ECL Exemplary Claim: 1  
DRWN 2 Drawing Figure(s); 2 Drawing Page(s)  
LN.CNT 2106  
CAS INDEXING IS AVAILABLE FOR THIS PATENT.  
AB This invention pertains to a method for liquid phase fluorination for perfluorination of a wide variety of hydrogen-containing compounds.  
IT 130085-03-9P  
(preparation of)  
RN 130085-03-9 USPATFULL  
CN 5,8,11,13,16,19-Hexaoxatricosane, 1,1,1,2,2,3,3,4,4,6,6,7,7,9,9,10,10,12,12,14,14,15,15,17,17,18,18,20,20,21,21,22,22,23,23,23-hexatriacontafluoro-

10/631,862

(9CI) (CA INDEX NAME)

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F<sub>3</sub>C-(CF<sub>2</sub>)<sub>3</sub>-O-CF<sub>2</sub>-CF<sub>2</sub>-O-CF<sub>2</sub>-CF<sub>2</sub>-O-CF<sub>2</sub>-O-CF<sub>2</sub>-CF<sub>2</sub>-O-CF<sub>2</sub>-

PAGE 1-B

-CF<sub>2</sub>-O-(CF<sub>2</sub>)<sub>3</sub>-CF<sub>3</sub>

L28 ANSWER 29 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 2

AN 1994:701682 CAPLUS

DN 121:301682

TI Liquid-phase fluorination

IN Bierschenk, Thomas R.; Juhlke, Timothy; Kawa, Hajimu; Lagow, Richard J.

PA Exfluor Research Corp., USA

SO U.S., 24 pp. Cont.-in-part of U.S. Ser. No. 822,637, abandoned.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 3

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 5332790	A	19940726	US 1993-28682	19930308
	US 5093432	A	19920303	US 1989-414119	19890928
	US 5322903	A	19940621	US 1992-823836	19920117
	US 5571870	A	19961105	US 1994-240225	19940510
	US 5674949	A	19971007	US 1995-466798	19950606
PRAI	US 1988-250376		19880928		
	US 1989-414119		19890928		
	US 1992-822637		19920117		
	US 1992-823836		19920117		
	US 1994-240225		19940510		

AB A method for replacing essentially all H atoms of H-containing compds. with F atoms comprises (a) continuously introducing a H-containing compound into a liquid

perfluorocarbon, perhalogenated chlorofluorocarbon or chloro fluoro ether medium while agitating the medium so that the H-containing compound is dissolved

or dispersed within the liquid medium; (b) introducing F gas diluted with an inert gas into the liquid medium without illumination with UV light to establish fluorination conditions wherein the liquid medium and F in the vapor space above the liquid medium do not form a flammable mixture; (c) continuing the introduction of F gas diluted with an inert gas until essentially all of the H atoms of the H-containing compound have been replaced with F atoms without substantial oligomerization or polymerization of the H-containing compound Perfluorinated acids (such as C<sub>7</sub>F<sub>15</sub>CO<sub>2</sub>H), perfluorinated polyethylene glycol and polypropylene glycol and their derivs. were prepared

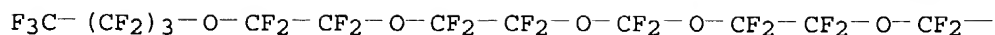
IT 130085-03-9P

RL: IMF (Industrial manufacture); PREP (Preparation)  
(liquid-phase perfluorination)

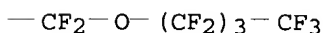
RN 130085-03-9 CAPLUS

CN 5,8,11,13,16,19-Hexaoxatricosane, 1,1,1,2,2,3,3,4,4,6,6,7,7,9,9,10,10,12,12,14,14,15,15,17,17,18,18,20,20,21,21,22,22,23,23,23-hexatriacontafluoro-  
(9CI) (CA INDEX NAME)

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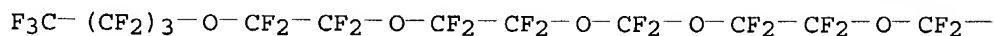


PAGE 1-B

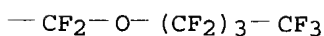


L28 ANSWER 30 OF 69 USPATFULL on STN  
 AN 94:53503 USPATFULL  
 TI Liquid-phase fluorination  
 IN Bierschenk, Thomas R., Round Rock, TX, United States  
 Juhlke, Timothy, Round Rock, TX, United States  
 Kawa, Hajimu, Austin, TX, United States  
 Lagow, Richard J., Austin, TX, United States  
 PA Exfluor Research Corporation, Austin, TX, United States (U.S. corporation)  
 PI US 5322903 19940621  
 AI US 1992-823836 19920117 (7)  
 RLI Continuation of Ser. No. US 1989-414119, filed on 28 Sep 1989, now patented, Pat. No. US 5093432, issued on 3 Mar 1992 which is a continuation-in-part of Ser. No. US 1988-250376, filed on 28 Sep 1988, now abandoned  
 DT Utility  
 FS Granted  
 EXNAM Primary Examiner: Krass, Frederick  
 LREP Hamilton, Brook, Smith & Reynolds  
 CLMN Number of Claims: 20  
 ECL Exemplary Claim: 1  
 DRWN 2 Drawing Figure(s); 2 Drawing Page(s)  
 LN.CNT 1950  
 CAS INDEXING IS AVAILABLE FOR THIS PATENT.  
 AB This invention pertains to a method for liquid phase fluorination for perfluorination of a wide variety of hydrogen-containing compounds.  
 IT 130085-03-9P  
 (preparation of)  
 RN 130085-03-9 USPATFULL  
 CN 5,8,11,13,16,19-Hexaoxatricosane, 1,1,1,2,2,3,3,4,4,6,6,7,7,9,9,10,10,12,12,14,14,15,15,17,17,18,18,20,20,21,21,22,22,23,23,23-hexatriacontafuoro-(9CI) (CA INDEX NAME)

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L28 ANSWER 36 OF 69 USPATFULL on STN  
 AN 92:46791 USPATFULL  
 TI Perfluorinated polyethers and process for their preparation  
 IN Kalota, Dennis J., Fenton, MO, United States

10/631,862

McConaghy, Jr., John S., St. Louis, MO, United States  
Foerst, Paul W., Chesterfield, MO, United States  
Liu, Paul H., Chesterfield, MO, United States  
Feher, Jr., Frank R., Belleville, IL, United States  
PA Monsanto Company, St. Louis, MO, United States (U.S. corporation)  
PI US 5120459 19920609  
AI US 1990-498055 19900323 (7)  
RLI Division of Ser. No. US 1989-150963, filed on 29 Jan 1989, now abandoned  
DT Utility  
FS Granted  
EXNAM Primary Examiner: McAvoy, Ellen  
LREP Brooks, W. W.  
CLMN Number of Claims: 2  
ECL Exemplary Claim: 1  
DRWN 2 Drawing Figure(s); 2 Drawing Page(s)  
LN.CNT 535  
CAS INDEXING IS AVAILABLE FOR THIS PATENT.  
AB Perfluorinated polyethers having the formula

$R_{\text{sub}f} O - (CF_{\text{sub}2} CF_{\text{sub}2} O)_{\text{sub}n} - R'_{\text{sub}f}$

wherein n is an integer of 1-11 and each of  $R_{\text{sub}f}$  and  $R'_{\text{sub}f}$  is a perfluorinated  $C_{\text{sub}1} - C_{\text{sub}5}$  -alkyl radical, dimers of such polyethers and carbon to carbon intramolecularly coupled cyclic derivatives of such polyethers are produced by direct fluorination of the polyethers in an inert solvent. Compositions of the perfluorinated polyethers and their derivatives are useful as functional fluids.

IT 125662-66-0P  
(manufacture of, solvent for)  
RN 125662-66-0 USPATFULL  
CN 2,5,8,11,14,17,20,23-Octaoxatetracosane, 1,1,1,3,3,4,4,6,6,7,7,9,9,10,10,12,12,13,13,15,15,16,16,18,18,19,19,21,21,22,22,24,24,24-tetratriacontafluoro- (9CI) (CA INDEX NAME)

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$F_3C - O - CF_2 - CF_2 - O - CF_2 - CF_2 - O - CF_2 - CF_2 - O - CF_2 - CF_2 - O - CF_2 - CF_2 -$

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$- O - CF_2 - CF_2 - O - CF_2 - CF_2 - O - CF_3$

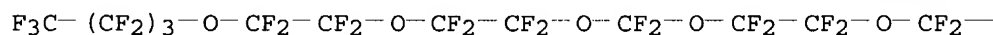
L28 ANSWER 37 OF 69 USPATFULL on STN  
AN 92:17225 USPATFULL  
TI Liquid phase fluorination  
IN Bierschenk, Thomas R., Round Rock, TX, United States  
Juhlke, Timothy, Round Rock, TX, United States  
Kawa, Hajimu, Austin, TX, United States  
Lagow, Richard J., Austin, TX, United States  
PA Exfluor Research Corporation, Austin, TX, United States (U.S. corporation)  
PI US 5093432 19920303  
AI US 1989-414119 19890928 (7)  
RLI Continuation-in-part of Ser. No. US 1988-250376, filed on 28 Sep 1988, now abandoned  
DT Utility  
FS Granted



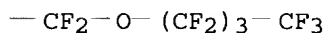
10/631,862

EXNAM Primary Examiner: Kight, III, John; Assistant Examiner: Krass, Frederick  
LREP Hamilton, Brook, Smith & Reynolds  
CLMN Number of Claims: 21  
ECL Exemplary Claim: 1  
DRWN 2 Drawing Figure(s); 2 Drawing Page(s)  
LN.CNT 2057  
CAS INDEXING IS AVAILABLE FOR THIS PATENT.  
AB This invention pertains to a method for liquid phase fluorination for  
perfluorination of a wide variety of hydrogen-containing compounds.  
IT 130085-03-9P  
(preparation of)  
RN 130085-03-9 USPATFULL  
CN 5,8,11,13,16,19-Hexaoxatricosane, 1,1,1,2,2,3,3,4,4,6,6,7,7,9,9,10,10,12,1  
2,14,14,15,15,17,17,18,18,20,20,21,21,22,22,23,23,23-hexatriacontafluoro-  
(9CI) (CA INDEX NAME)

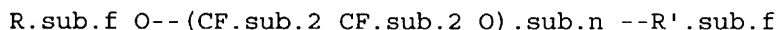
PAGE 1-A



PAGE 1-B



L28 ANSWER 38 OF 69 USPATFULL on STN  
AN 91:106001 USPATFULL  
TI Novel perfluorinated polyethers and process for their preparation  
IN Kalota, Dennis J., Fenton, MO, United States  
McConaghy, Jr., John S., St. Louis, MO, United States  
Foerst, Paul W., Chesterfield, MO, United States  
Liu, Paul H., Chesterfield, MO, United States  
Feher, Jr., Frank R., Belleville, IL, United States  
PA Monsanto Company, St. Louis, MO, United States (U.S. corporation)  
PI US 5076949 19911231  
AI US 1990-498124 19900323 (7)  
RLI Division of Ser. No. US 1989-150963, filed on 29 Jan 1989, now abandoned  
DT Utility  
FS Granted  
EXNAM Primary Examiner: Willis, Jr., Prince; Assistant Examiner: McAvoy, Ellen  
LREP Brooks. W. W.  
CLMN Number of Claims: 5  
ECL Exemplary Claim: 1  
DRWN 2 Drawing Figure(s); 2 Drawing Page(s)  
LN.CNT 549  
CAS INDEXING IS AVAILABLE FOR THIS PATENT.  
AB Perfluorinated polyethers having the formula



wherein n is an integer of 1-11 and each of R.sub.f and R'.sub.f is a  
perfluorinated C.sub.1 -C.sub.5 -alkyl radical, dimers of such  
polyethers and carbon to carbon intramolecularly coupled cyclic  
derivatives of such polyethers are produced by direct fluorination of  
the polyethers in an inert solvent. Compositions of the perfluorinated  
polyethers and their derivatives are useful as functional fluids.

IT 125662-66-0P  
(manufacture of, solvent for)

10/631,862

RN 125662-66-0 USPATFULL  
CN 2,5,8,11,14,17,20,23-Octaoxatetracosane, 1,1,1,3,3,4,4,6,6,7,7,9,9,10,10,1  
2,12,13,13,15,15,16,16,18,18,19,19,21,21,22,22,24,24,24-  
tetratriacontafluoro- (9CI) (CA INDEX NAME)

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$\text{F}_3\text{C}-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-$

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$-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_3$

L28 ANSWER 39 OF 69 USPATFULL on STN  
AN 91:17276 USPATFULL  
TI Novel perfluorinated polyethers and process for their preparation  
IN Kalota, Dennis J., Fenton, MO, United States  
McConaghy, Jr., John S., St. Louis, MO, United States  
Foerst, Paul W., Chesterfield, MO, United States  
Liu, Paul H., Chesterfield, MO, United States  
Feher, Jr., Frank R., Belleville, IL, United States  
PA Monsanto Company, St. Louis, MO, United States (U.S. corporation)  
PI US 4996369 19910226  
AI US 1990-498057 19900523 (7)  
RLI Division of Ser. No. US 1989-150963, filed on 29 Jan 1989  
DT Utility  
FS Granted  
EXNAM Primary Examiner: Mars, Howard T.  
LREP Brooks, W.  
CLMN Number of Claims: 1  
ECL Exemplary Claim: 1  
DRWN 2 Drawing Figure(s); 2 Drawing Page(s)  
LN.CNT 533  
CAS INDEXING IS AVAILABLE FOR THIS PATENT.  
AB Perfluorinated polyethers having the formula

$\text{R.sub.f O}--(\text{CF.sub.2 CF.sub.2 O}).\text{sub.n} --\text{R'}.sub.f$

wherein n is and integer of 1-11 and each of R.sub.f and R'.sub.f is a perfluorinated C.sub.1 -C.sub.5 -alkyl radical, dimers of such polyethers and carbon to carbon intramolecularly coupled cyclic derivatives of such polyethers are produced by direct fluorination of the polyethers in an inert solvent. Compositions of the perfluorinated polyethers and their derivatives are useful as functional fluids.

IT 125662-66-0P  
(manufacture of, solvent for)  
RN 125662-66-0 USPATFULL  
CN 2,5,8,11,14,17,20,23-Octaoxatetracosane, 1,1,1,3,3,4,4,6,6,7,7,9,9,10,10,1  
2,12,13,13,15,15,16,16,18,18,19,19,21,21,22,22,24,24,24-  
tetratriacontafluoro- (9CI) (CA INDEX NAME)

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$\text{F}_3\text{C}-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-$

—O—CF<sub>2</sub>—CF<sub>2</sub>—O—CF<sub>2</sub>—CF<sub>2</sub>—O—CF<sub>3</sub>

L28 ANSWER 44 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1990:121089 CAPLUS

DN 112:121089

TI Perfluorination of alcohol ethoxylates

IN Feher, Frank Ronald; Foerst, Paul Wayne; Liu, Paul Ho; Kalota, Dennis  
Jerome; McConaghy, John Stead, Jr.

PA Monsanto Co., USA

SO Eur. Pat. Appl., 12 pp.

CODEN: EPXXDW

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	EP 332601	A1	19890913	EP 1989-870017	19890127
	R: AT, BE, CH, DE, ES, FR, GB, GR, IT, LI, LU, NL, SE				
	AU 8928869	A1	19890803	AU 1989-28869	19890127
	AU 607579	B2	19910307		
	JP 01225628	A2	19890908	JP 1989-19415	19890127
	US 5076949	A	19911231	US 1990-498124	19900323
	US 5120459	A	19920609	US 1990-498055	19900323
	US 4996369	A	19910226	US 1990-498057	19900523
	JP 06025404	A2	19940201	JP 1993-115041	19930517
	JP 07086139	B4	19950920		
	JP 06025405	A2	19940201	JP 1993-115042	19930517
	JP 07086140	B4	19950920		
	JP 06080773	A2	19940322	JP 1993-115043	19930517
	JP 07088423	B4	19950927		

PRAI US 1988-150963 19880129

AB The title compds. RO(CF<sub>2</sub>CF<sub>2</sub>O)<sub>n</sub>R<sub>1</sub> (R, R<sub>1</sub> = perfluorinated C<sub>1</sub>-5 alkyl; n 1-11) are prepared by reacting F(g) with alc. ethoxylates R<sub>2</sub>O(CH<sub>2</sub>CH<sub>2</sub>O)<sub>n</sub>R<sub>3</sub> (R<sub>2</sub>, R<sub>3</sub> = C<sub>1</sub>-5 alkyl; n = 1-11) in an inert solvent and separating the product. A process schematic and a reactor diagram are presented. Thus, 250 g heptaglyme and a slurry of 1110 g NaF in 4 L 1,1,2-trichloro-1,2,2-trifluoroethane was charged into a stirred (1200 rpm) reactor, a mixture of F and N added at 15-25° for 4 h, 7105 g of the fluorinated oil intermediate (5-10% H content) was charged into a reactor with 300 g NaF, the oil reacted with F at 31-128° for 177 min, the treated oil reacted with F at 29-253° for 230 min, and distilled to give a title product having average mol. weight 1000, b.p. (760 torr) 215°, pour point -25°, and d<sub>20</sub> 1.72 g/mL. The distillation bottoms contained perfluoroheptaglyme dimer and oligomers having average mol. weight 1900, b.p. 200°/4 torr, pour point -70°, and d<sub>20</sub> 1.81 g/mL.

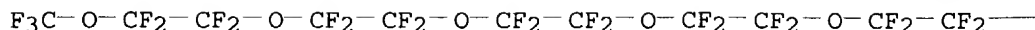
IT 125662-66-0P

RL: IMF (Industrial manufacture); PREP (Preparation)  
(manufacture of, solvent for)

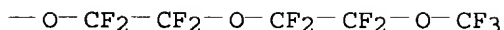
RN 125662-66-0 CAPLUS

CN 2,5,8,11,14,17,20,23-Octaoxatetracosane, 1,1,1,3,3,4,4,6,6,7,7,9,9,10,10,1  
2,12,13,13,15,15,16,16,18,18,19,19,21,21,22,22,24,24,24-  
tetratriacontafluoro- (9CI) (CA INDEX NAME)

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L28 ANSWER 45 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 4

AN 1989:438876 CAPLUS

DN 111:38876

TI Preparation of carbonyl fluoride compounds

IN Okabe, Jun; Tatsu, Haruyoshi

PA Nippon Mectron Co., Ltd., Japan

SO U.S., 7 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 4769184	A	19880906	US 1987-121135	19871116
	JP 01066139	A2	19890313	JP 1987-222946	19870908
	JP 08019035	B4	19960228		
	JP 01093557	A2	19890412	JP 1987-249588	19871002
	JP 2726824	B2	19980311		
PRAI	JP 1987-222946		19870908		
	JP 1987-249588		19871002		

OS MARPAT 111:38876

AB A process for producing XCOF (I; X = F, CF<sub>3</sub>) or I (X = CF<sub>3</sub>CF<sub>2</sub>), useful as intermediates for producing perfluoro(alkyl vinyl ethers) which are monomers for producing F-containing resins, F-containing rubber, etc., comprised

thermally decomposing RfO(CF<sub>2</sub>CF<sub>2</sub>O)<sub>a</sub>(CF<sub>2</sub>O)<sub>b</sub>(O)cRf' (Rf = perfluoroalkyl; Rf' = COF, CF<sub>3</sub>; the CF<sub>2</sub>O and O groups are distributed at random; a, b ≠ 0; c can be 0; a + b + c ≤ .apprx.200) or RfO(CFXCF<sub>2</sub>O)<sub>n</sub>CFX'Y (X' = CF<sub>3</sub>, F, H; Y = COF, CO<sub>2</sub>H, CO<sub>2</sub>R, CF<sub>3</sub>; R = alkyl; n = 1-50), resp. F<sub>2</sub>C:CF<sub>2</sub> and O<sub>2</sub> were irradiated with UV to give F<sub>3</sub>CO(CF<sub>2</sub>OF<sub>2</sub>O)<sub>8</sub>(CF<sub>2</sub>O)<sub>2400.4</sub>COF, thermal decomposition of which at 200° over activated C gave a mixture of 78.2% COF<sub>2</sub> and 21.8% F<sub>3</sub>CCOF. I (X = F, CF<sub>3</sub>) so produced contain no Cl-based impurities.

IT 119214-96-9P

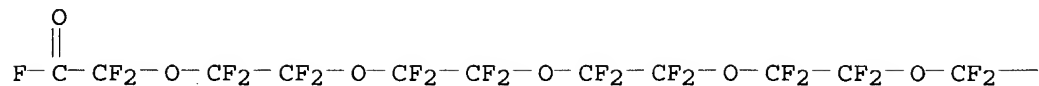
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and reaction of, in synthesis of carbonyl fluorides)

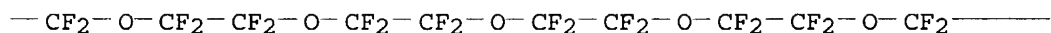
RN 119214-96-9 CAPLUS

CN 3,6,9,12,15,18,21,24,27,30,33,36,39,42,45,48,51,54,57,60,63,66,69-  
Tricosaoxahenheptacontanoyl fluoride, 2,2,4,4,5,5,7,7,8,8,10,10,11,11,13,13,14,14,16,16,17,17,19,19,20,20,22,22,23,23,25,25,26,26,28,28,29,29,31,31,32,32,34,34,35,35,37,37,38,38,40,40,41,41,43,43,44,44,46,46,47,47,49,49,50,50,52,52,53,53,55,55,56,56,58,58,59,59,61,61,62,62,64,64,65,65,67,67,68,68,70,70,71,71,71-pentanonacontafluoro- (9CI) (CA INDEX NAME)

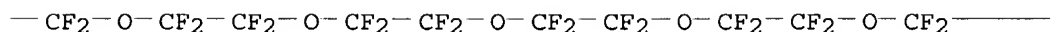
PAGE 1-A



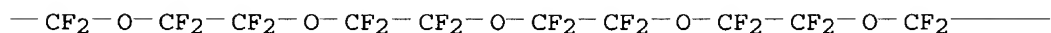
PAGE 1-B



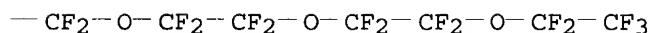
PAGE 1-C



PAGE 1-D



PAGE 1-E



L28 ANSWER 53 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 7

AN 1986:5566 CAPLUS

DN 104:5566

TI Perfluorocarbon ethers from a high-molecular-weight polyether

IN Lagow, Richard J.; Gerhardt, Glenn E.

PA University of Texas, USA

SO U.S., 16 pp. Cont. U.S. Ser. No. 139,181 abandoned.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 4523039	A	19850611	US 1983-563013	19831219
PRAI	US 1978-901905		19780501		
	US 1980-139181		19800411		

AB Fluorocarbon ethers were prepared by fluorination of a high mol. weight polyether with F<sub>2</sub> to produce a fluorinated polyether, which was depolymd. by further treatment with F<sub>2</sub> at 55-210°. Thus, polyethylene oxide

was fluoroinated with flowing F<sub>2</sub>-He, using LaMar techniques, at ambient temperature for 12 days, at 90° for 2 days, and at 110° for 7 days, to give a mixture of compds. including CF<sub>4</sub>, COF<sub>2</sub>, F<sub>3</sub>CO(CF<sub>2</sub>CF<sub>2</sub>O)<sub>n</sub>R (n = 1-6; R = CF<sub>3</sub>, C<sub>2</sub>F<sub>5</sub>) (all permutations), and C<sub>2</sub>F<sub>5</sub>O(CF<sub>2</sub>CF<sub>2</sub>O)<sub>m</sub>C<sub>2</sub>F<sub>5</sub> (m = 1-3), which were separated and characterized by IR, <sup>19</sup>F NMR, and mass spectroscopy.

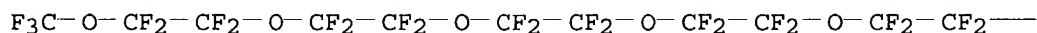
IT 64028-08-6P

RL: SPN (Synthetic preparation); PREP (Preparation  
(preparation of, by fluorination-depolymn. of polyether, and spectral  
characterization of)

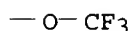
RN 64028-08-6 CAPLUS

CN 2,5,8,11,14,17-Hexaoxaoctadecane, 1,1,1,2,2,4,4,6,6,7,7,9,9,10,10,12,12,13,13,15,15,16,16,18,18,18-hexacosafuoro- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 1-B



L28 ANSWER 65 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1977:517527 CAPLUS

DN 87:117527

TI Synthesis of perfluoropoly(ethylene glycol) ethers  $\text{CF}_3[\text{OCF}_2\text{CF}_2]_n\text{ORf}$  ( $\text{Rf} = \text{CF}_3$  or  $\text{C}_2\text{F}_5$ ;  $n = 1-5$ )

AU     Gerhardt, Glenn E.; Laqow, Richard J.

CS Dep. Chem., Univ. Texas, Austin, TX, USA

S0 Journal of the Chemical Society, Chemical Communications (1977), (8), 259-60

CODEN: JCCCAT; ISSN: 0022-4936

DT Journal

LA English

AB Finely ground (<120 mesh) poly(ethylene oxide) reacted with elemental F, under conditions carefully regulated to fragment and perfluorinate the polyether system, to give  $\text{CF}_3[\text{O}(\text{CF}_2)_2]_n\text{OR}$  ( $\text{R} = \text{CF}_3, \text{C}_2\text{F}_5, n = 1-5$ ).

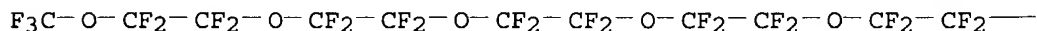
IT 64028-08-6P

RL: SPN (Synthetic preparation); PREP (Preparation)  
(preparation of)

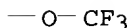
RN 64028-08-6 CAPLUS

CN 2,5,8,11,14,17-Hexaoxaoctadecane, 1,1,1,2,2,4,4,6,6,7,7,9,9,10,10,12,12,13,13,15,15,16,16,18,18,18-hexacosafuoro- (9CI) (CA INDEX NAME)

PAGE 1-A



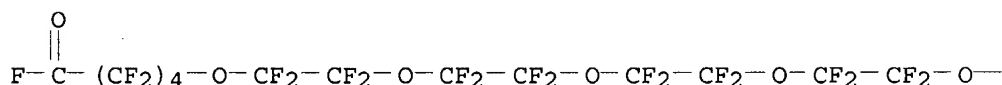
PAGE 1-B



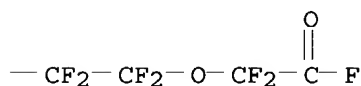
10/631,862

AN 1976:493374 CAPLUS  
DN 85:93374  
TI The addition of tetrafluoroethylene oxide to F-glutaryl fluoride.  
Relative reactivities of acid fluorides  
AU Anderson, R.; Baucom, K. B.; Psarras, T.; Snyder, C. E.; Cochoy, R. E.  
CS PCR, Inc., Gainesville, FL, USA  
SO Journal of Fluorine Chemistry (1976), 7(6), 581-8  
CODEN: JFLCAR; ISSN: 0022-1139  
DT Journal  
LA English  
AB Data obtained from the addition of tetrafluoroethylene oxide to F-glutaryl fluoride [FOC(CF<sub>2</sub>)<sub>3</sub>COF] show significant differences to exist between the relative reactivities of the acid fluoride groups involved. The order of reactivity is F-glutaryl fluoride > -OCF<sub>2</sub>COF > -CF<sub>2</sub>CF<sub>2</sub>COF.  
IT 60127-05-1P  
RL: SPN (Synthetic preparation); PREP (Preparation)  
(preparation of)  
RN 60127-05-1 CAPLUS  
CN 3,6,9,12,15,18-Hexaoxatricosanedioyl difluoride,  
2,2,4,4,5,5,7,7,8,8,10,10,11,11,13,13,14,14,16,16,17,17,19,19,20,20,21,21,  
22,22-triacontafuoro- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 1-B



L28 ANSWER 67 OF 69 CAPLUS COPYRIGHT 2004 ACS on STN  
AN 1974:26718 CAPLUS  
DN 80:26718  
TI Nature of the addition of perfluoroolefin oxides to perfluorodicarboxylic acid difluorides  
AU Skoblikova, V. I.; Sass, V. P.; Ershov, A. E.; Senyushov, L. N.; Sokolov, L. F.; Berenblit, V. V.; Sokolov, S. V.  
CS Vses. Nauchno-Issled Inst. Sint. Kauch., Leningrad, USSR  
SO Zhurnal Organicheskoi Khimii (1973), 9(10), 2021-5  
CODEN: ZORKAE; ISSN: 0514-7492  
DT Journal  
LA Russian  
AB FCOCF<sub>2</sub>COF (I) reacted with tetrafluoroethylene oxide (II) in diglyme containing CsF at -25° to give mixts. containing MeO<sub>2</sub>CCF<sub>2</sub>(CF<sub>2</sub>OCF<sub>2</sub>)<sub>n</sub>CO<sub>2</sub>Me (n = 1-6) after quenching with MeOH; FCO(CF<sub>2</sub>)<sub>4</sub>COF reacted analogously with II to give MeO<sub>2</sub>C(CF<sub>2</sub>)<sub>4</sub>(CF<sub>2</sub>OCF<sub>2</sub>)<sub>n</sub>CO<sub>2</sub>Me (n = 1-3). Reaction of I with hexafluoropropylene oxide afforded products of addition at both carbonyl groups of I, i.e., MeO<sub>2</sub>C[CF(CF<sub>3</sub>)OCF<sub>2</sub>]<sub>m</sub>CF<sub>2</sub>[CF<sub>2</sub>OCF(CF<sub>3</sub>)]<sub>n</sub>CO<sub>2</sub>Me (m = n = 1,2; m = 1-3, n = m-1).  
IT 50733-65-8P  
RL: SPN (Synthetic preparation); PREP (Preparation)  
(preparation of)  
RN 50733-65-8 CAPLUS

CN 3,6,9,12,15,18-Hexaoxaheneicosanedioic acid, 2,2,4,4,5,5,7,7,8,8,10,10,11,11,13,13,14,14,16,16,17,17,19,19,20,20-hexacosafuoro-, dimethyl ester  
(9CI) (CA INDEX NAME)

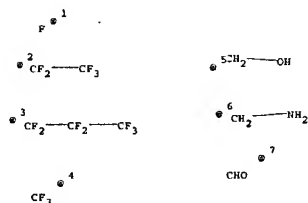
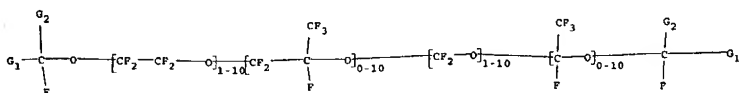
$$\text{MeO}-\overset{\text{O}}{\parallel}{\text{C}}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-\text{O}-\text{CF}_2-\text{CF}_2-$$
$$\text{---O---CF}_2\text{---CF}_2\text{---O---CF}_2\text{---C(=O)OMe}$$

-14.70

STN INTERNATIONAL SESSION SUSPENDED AT 07:38:59 ON 10 NOV 2004



Structure 10630698 to large  
to search



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 21 22 23 25 26 27 28 29 30 31 32 33  
34 35 36 37 38 39 40 41 42 43 45

chain bonds :

2-7 3-4 4-6 8-10 9-11 21-22 22-23 22-25 22-26 26-27 27-28 28-29 29-30 30-31  
31-32 31-33 31-34 34-35 35-36 36-37 37-40 37-38 37-39 40-41 41-42 41-43  
41-45

exact/norm bonds :

21-22 22-25 22-26 31-34 36-37 37-40 40-41 41-43 41-45

exact bonds :

2-7 3-4 4-6 8-10 9-11 22-23 26-27 27-28 28-29 29-30 30-31 31-32 31-33 34-35  
35-36 37-38 37-39 41-42

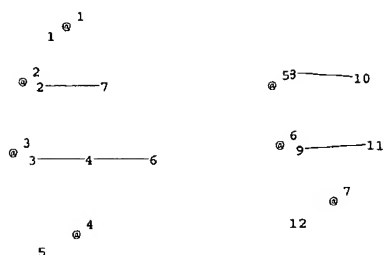
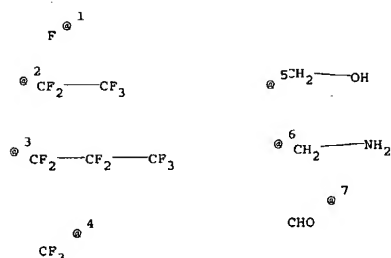
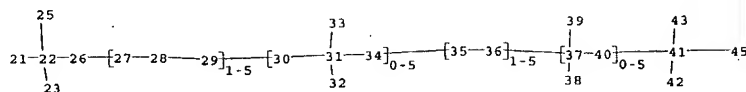
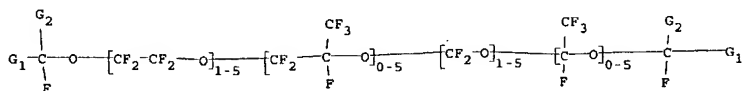
G1:[\*1],[\*2],[\*3],[\*4],[\*5],[\*6],[\*7]

G2:[\*1],[\*2]

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS  
10:CLASS 11:CLASS 12:CLASS 21:CLASS 22:CLASS 23:CLASS 25:CLASS 26:CLASS 27:CLASS  
28:CLASS 29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS 35:CLASS 36:CLASS  
37:CLASS 38:CLASS 39:CLASS 40:CLASS 41:CLASS 42:CLASS 43:CLASS 45:CLASS

structure too large to search



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 21 22 23 25 26 27 28 29 30 31 32 33  
34 35 36 37 38 39 40 41 42 43 45

chain bonds :

2-7 3-4 4-6 8-10 9-11 21-22 22-23 22-25 22-26 26-27 27-28 28-29 29-30 30-31  
31-32 31-33 31-34 34-35 35-36 36-37 37-40 37-38 37-39 40-41 41-42 41-43  
41-45

exact/norm bonds :

21-22 22-25 22-26 31-34 36-37 37-40 40-41 41-43 41-45

exact bonds :

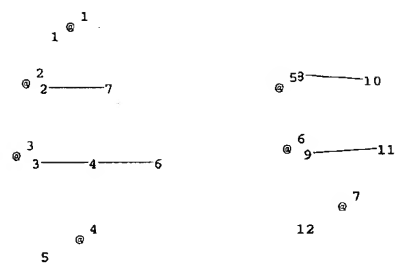
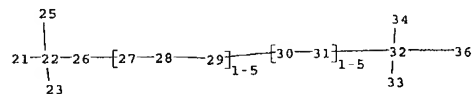
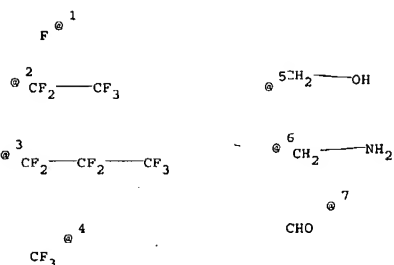
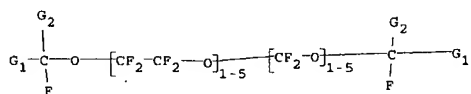
2-7 3-4 4-6 8-10 9-11 22-23 26-27 27-28 28-29 29-30 30-31 31-32 31-33 34-35  
35-36 37-38 37-39 41-42

G1: [\*1], [\*2], [\*3], [\*4], [\*5], [\*6], [\*7]

G2: [\*1], [\*2]

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS  
10:CLASS 11:CLASS 12:CLASS 21:CLASS 22:CLASS 23:CLASS 25:CLASS 26:CLASS 27:CLASS  
28:CLASS 29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS 35:CLASS 36:CLASS  
37:CLASS 38:CLASS 39:CLASS 40:CLASS 41:CLASS 42:CLASS 43:CLASS 45:CLASS



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 21 22 23 25 26 27 28 29 30 31 32 33  
34 36

chain bonds :

2-7 3-4 4-6 8-10 9-11 21-22 22-23 22-25 22-26 26-27 27-28 28-29 29-30 30-31  
31-32 32-36 32-33 32-34

exact/norm bonds :

21-22 22-25 22-26 31-32 32-36 32-34

exact bonds :

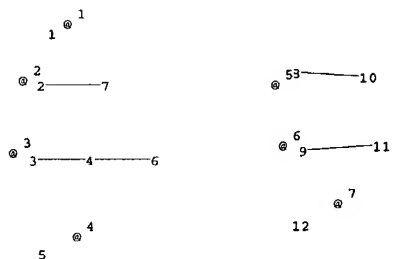
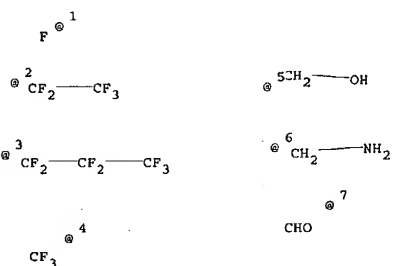
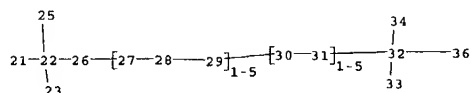
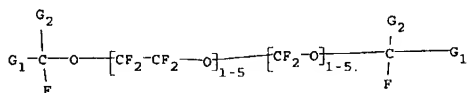
2-7 3-4 4-6 8-10 9-11 22-23 26-27 27-28 28-29 29-30 30-31 32-33

G1:[\*1],[\*2],[\*3],[\*4],[\*5],[\*6],[\*7]

G2:[\*1],[\*2]

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS  
10:CLASS 11:CLASS 12:CLASS 21:CLASS 22:CLASS 23:CLASS 25:CLASS 26:CLASS 27:CLASS  
28:CLASS 29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS 36:CLASS



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 21 22 23 25 26 27 28 29 30 31 32 33  
34 36

chain bonds :

2-7 3-4 4-6 8-10 9-11 21-22 22-23 22-25 22-26 26-27 27-28 28-29 29-30 30-31  
31-32 32-36 32-33 32-34

exact/norm bonds :

21-22 22-25 22-26 31-32 32-36 32-34

exact bonds :

2-7 3-4 4-6 8-10 9-11 22-23 26-27 27-28 28-29 29-30 30-31 32-33

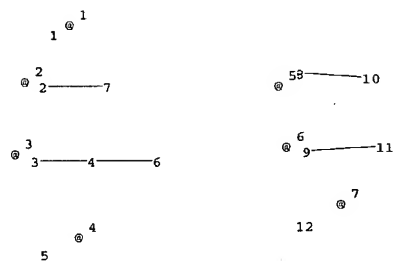
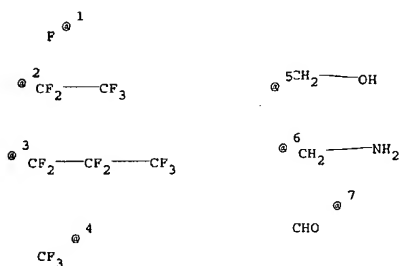
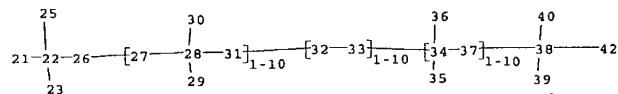
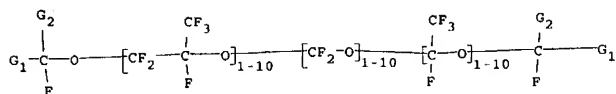
1:[\*1],[\*2],[\*3],[\*4],[\*5],[\*6],[\*7]

2:[\*1],[\*2]

atch level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS  
10:CLASS 11:CLASS 12:CLASS 21:CLASS 22:CLASS 23:CLASS 25:CLASS 26:CLASS 27:CLASS  
28:CLASS 29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS 36:CLASS

Structure too large to  
search



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 21 22 23 25 26 27 28 29 30 31 32 33  
34 35 36 37 38 39 40 42

chain bonds :

2-7 3-4 4-6 8-10 9-11 21-22 22-23 22-25 22-26 26-27 27-28 28-29 28-30 28-31  
31-32 32-33 33-34 34-37 34-35 34-36 37-38 38-39 38-40 38-42

exact/norm bonds :

21-22 22-25 22-26 28-31 33-34 34-37 37-38 38-40 38-42

exact bonds :

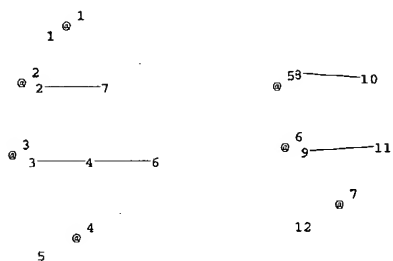
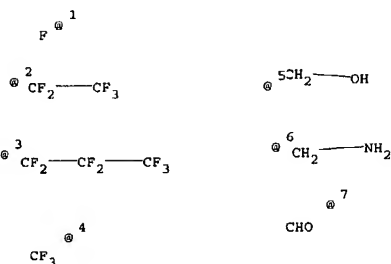
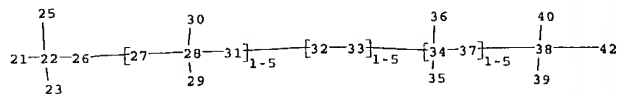
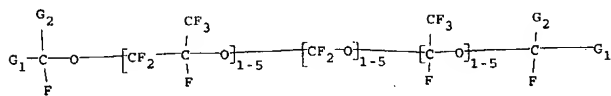
2-7 3-4 4-6 8-10 9-11 22-23 26-27 27-28 28-29 28-30 31-32 32-33 34-35 34-36  
38-39

G1:[\*1],[\*2],[\*3],[\*4],[\*5],[\*6],[\*7]

G2:[\*1],[\*2]

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS  
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28:CLASS 29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS 35:CLASS 36:CLASS  
37:CLASS 38:CLASS 39:CLASS 40:CLASS 42:CLASS



chain nodes :

1 2 3 4 5 6 7 8 9 10 11 12 21 22 23 25 26 27 28 29 30 31 32 33  
34 35 36 37 38 39 40 42

chain bonds :

2-7 3-4 4-6 8-10 9-11 21-22 22-23 22-25 22-26 26-27 27-28 28-29 28-30 28-31  
31-32 32-33 33-34 34-37 34-35 34-36 37-38 38-39 38-40 38-42

exact/norm bonds :

21-22 22-25 22-26 28-31 33-34 34-37 37-38 38-40 38-42

exact bonds :

2-7 3-4 4-6 8-10 9-11 22-23 26-27 27-28 28-29 28-30 31-32 32-33 34-35 34-36  
38-39

G1:[\*1],[\*2],[\*3],[\*4],[\*5],[\*6],[\*7]

G2:[\*1],[\*2]

Match level :

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37:CLASS 38:CLASS 39:CLASS 40:CLASS 42:CLASS